

The Genetic Code and Enzyme Formation

Introduction

DNA is composed of two long strands of repeating nucleotide bases (A, T, C, G). You have already learned how DNA is able to replicate itself, but how is this simple molecule responsible for all of the traits that exist in every living organism?

The flow of information in organisms, from DNA to traits, is called the **Central Dogma**. Today, you will use paper models to show this important concept.

Procedure

As you work through the procedure, there will be questions asked at certain points. Record the answers to these questions on your **Genetic Code and Enzyme Formation Answer Sheet** before moving on to the next step.

Part 1: Transcription (DNA → mRNA)

Lay out your photocopy of "The Cell". Notice that it shows a DNA molecule within the nucleus of the cell. The line dividing the page is the nuclear membrane. The cytoplasm is on the right side of the page; imagine that the cytoplasm extends off the page and on to your table.

Place the 3 **Transfer RNA** molecules (**tRNA**) and the 4 **amino acid** molecules in random arrangements on your table (the cytoplasm of the cell).

The remaining molecule, **mRNA**, is called a **messenger RNA**. Place the mRNA molecule in the nucleus of the cell next to the DNA molecule. Compare the DNA and mRNA molecules.

1. How many strands of DNA are there compared to mRNA?
2. What nucleotide base is found on mRNA, but not DNA? What nucleotide base of DNA is this replacing?
3. Is this nucleotide base of mRNA also found on the tRNA?

In an RNA molecule, the thymine (T) nucleotide in DNA is replaced by the uracil (U) nucleotide base. This means that the U nucleotide will bond with the A nucleotide in mRNA and tRNA (as opposed to the A-T bond found in DNA).

The DNA molecule in the nucleus of your paper cell has a "code" written down on one of its strands. The letters used in this code are the sequences of A, T, C, and G nucleotides. In this paper model, this code is the gene for brown eyes (in real DNA, the gene for brown eyes would be thousands of base pairs long).

When this code reaches the cytoplasm, it will give instructions on how to make **enzyme B**, the enzyme needed to make brown eyes. This code needs to somehow get from the DNA to the cytoplasm outside of the nucleus. A helper molecule, mRNA (messenger RNA), accomplishes this task by carrying the message of the code from the DNA to the cytoplasm.

Slide your mRNA molecule next to your DNA molecule. Every nucleotide base should fit next to each other.

4. What is the role of mRNA in the cell?

Genetic Code And Enzyme Formation Answer

Judd E. Hollander



Genetic Code And Enzyme Formation Answer:

Diploma in pharmacy Exit Exam Book Mr. Chandan Kumar Raikwar, Mr. Sachin Kumar Agrahari, Dr. Ravi Dnyandeo, Ms. Sandhya Ramesh Kolekar, Ms. Nikita Bibhishan Durande, 2024-10-14 This comprehensive MCQ book is meticulously designed to assist Diploma in Pharmacy students in their preparation for the Exit Exam The book covers all essential subjects ensuring a thorough understanding of key concepts Each chapter includes a diverse set of multiple choice questions providing students with a robust tool to assess their knowledge and readiness for the exam

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TEXT BOOK OF BIOCHEMISTRY Prof. (Dr.) Bhoomika, Dr. Sunita S. Pachori, Abhilasha Gupta, Kuldeep Saini, Mr. Shashank Chaurasiya, 2025-09-19 The Textbook of Biochemistry provides a comprehensive understanding of the molecular basis of life systematically covering the fundamental biomolecules and their roles in maintaining biological functions Beginning with biomolecules the book introduces carbohydrates lipids nucleic acids amino acids and proteins highlighting their classification chemical nature and biological significance The section on bioenergetics explores the concepts of free energy enthalpy and entropy distinguishing between endergonic and exergonic reactions and explaining redox potential energy rich compounds and the significance of ATP and cyclic AMP Carbohydrate metabolism is detailed with pathways such as glycolysis the citric acid cycle HMP shunt gluconeogenesis and glycogen metabolism along with clinical conditions like G6PD deficiency glycogen storage diseases and diabetes mellitus The discussion on biological oxidation emphasizes the electron transport chain oxidative phosphorylation substrate level phosphorylation and the effects of inhibitors and uncouplers Lipid metabolism is explained through the oxidation of fatty acids ketone body metabolism de novo fatty acid synthesis and the biological significance of cholesterol including its conversion into bile acids steroid hormones and vitamin D with attention to disorders such as hypercholesterolemia atherosclerosis fatty liver and obesity Amino acid metabolism covers transamination deamination decarboxylation the urea cycle and its disorders catabolism of phenylalanine and tyrosine and their related diseases such as phenylketonuria albinism alkaptonuria and tyrosinemia It also highlights the synthesis and significance of key biological molecules such as dopamine serotonin 5 HT melatonin noradrenaline and adrenaline alongside heme

catabolism hyperbilirubinemia and jaundice The section on nucleic acid metabolism and genetic information transfer describes the biosynthesis and catabolism of purine and pyrimidine nucleotides clinical conditions like gout the structural organization of DNA and RNA DNA replication transcription the genetic code translation and inhibitors of protein synthesis The chapter on enzymes introduces their properties nomenclature and IUB classification followed by enzyme kinetics explained through Michaelis and Lineweaver Burke plots It also addresses enzyme inhibition with examples regulatory mechanisms including induction repression and allosteric regulation and their therapeutic and diagnostic applications Isoenzymes are discussed in medical contexts while the section on coenzymes explains their structures and biochemical functions Altogether this textbook integrates biochemical principles with clinical correlations making it an essential guide for pharmacy medical and life science students bridging molecular understanding with disease mechanisms and therapeutic applications

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Biochemistry Francis J. Chlapowski,1993 Designed to provide medical students with an instrument for self assessment and review in the field of biochemistry It contains 500 multiple choice questions designed to parallel the questions contained in Step 1 of the United States Medical Licensure Examination USMLE

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Microbiology Abstracts ,1991-10 *Structure and Function in Man* Stanley Wallace Jacob,Clarice Ashworth Francone,1970

Introduction to General, Organic, and Biological Chemistry Martha J. Gilleland,1982

Dictionary of Biochemistry and Molecular Biology J. Stenesh,1989-10-20 In response to the expansion of knowledge in biochemistry and molecular biology the Second Edition of this reference has been completely revised and updated with approximately 16 000 new entries Names of specific compounds and other substances have been substantially enlarged and definitions have been expanded for clarity and precision Information is drawn from over 500 books and 1 000 articles including recommendations of the Commission on Biochemical Nomenclature the International Union of Pure and Applied Chemistry and the International Union of Biochemistry Terms used by biochemists from a broad range of sciences such as chemistry immunology genetics virology biophysics and microbiology are included Abbreviations both standard and nonstandard are also provided as well as cross referenced synonymous expressions

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pressure drop of example 1 was 1 psig find the proper size tubing step 1

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