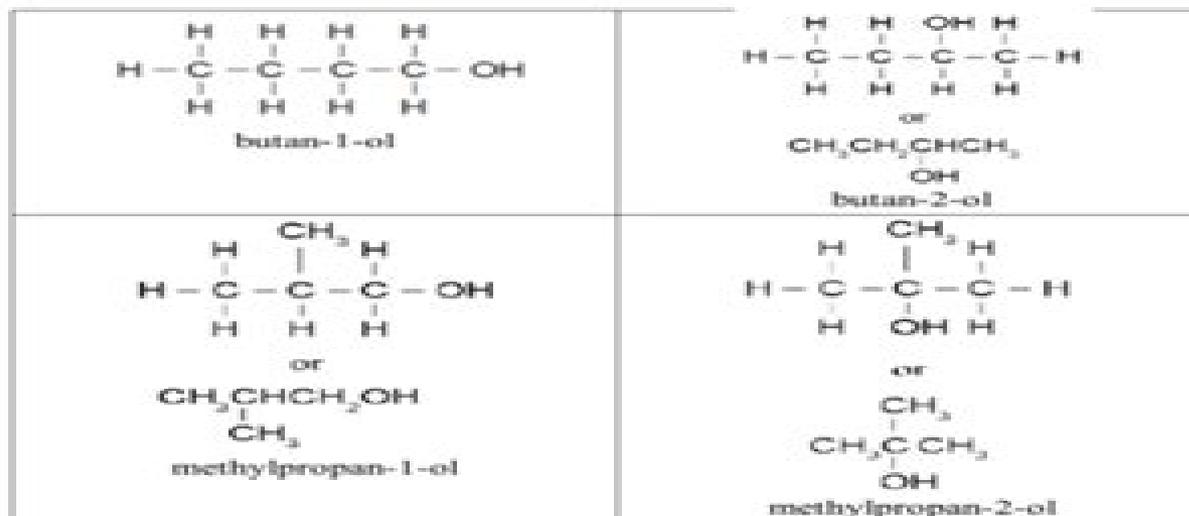


ANSWERS: Isomers

1)



2) No; for a molecule to exist as geometric isomers, it must contain a double bond, and each carbon (involved in the double bond) must have two different atoms / groups attached to it. But-1-ene has a double bond, but the atoms attached to one carbon are both the same (two hydrogen atoms) so it does not form a geometric isomer.

3) a)



b)



c) Structural isomers are compounds with the same molecular formula (they have the same type and number of atoms) but different structural formulae (structures).

$\text{C}_2\text{H}_4\text{Cl}_2$ has two structural isomers, one has a condensed structural formula $\text{CH}_2\text{Cl}-\text{CH}_2\text{Cl}$ and the other $\text{CH}_3-\text{CHCl}_2$

$\text{C}_2\text{H}_2\text{Cl}_2$ also has structural isomers, one has a condensed structural formula $\text{CH}_2\text{Cl}-\text{CHCl}$ and the other CH_2-CCl_2

Geometric Isomers Lab Answers

**Harold R. Hunt, Toby F. Block, George
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