

Grade 7

Natural Sciences

Term 4

Exam paper 1

Section	Questions	Answers
Section A Name 3 elements: _____ Instructions: 1. Label three (3) different water states. 2. Name an alloy (two different) and explain it. 3. Describe the structure of diamond and graphite. 4. Write clearly and legibly, and show all the possible chemical equations for: 	QUESTION 1 Name the three natural sources for water. (3) Give water one natural use and one application. (2) 1.1 What is the best description for the: 1.1.1 Earth's atmosphere (individuals) (1) 1.1.2 The Earth's crust (individuals) (1) 1.1.3 The Earth's interior (individuals) (1) 1.2 What is the effect of increasing atmospheric CO ₂ ? (1) 1.3 A hydrocarbon found in the atmosphere is an important raw material for the production of a plastic called poly(ethene). 1.3.1 Write the chemical formula for ethene. (1) 1.3.2 Why do plastic bottles often become brittle? (1) 1.3.3 To increase the length of plastic in a product, what must the conditions of polymerisation be? (1) 1.4 What is the main cooling mechanism for the Earth? 1.4.1 Name the process. (1) 1.4.2 Explain how the process works. (1) 1.5 What is the main source of energy for the Earth? 1.5.1 Name the source. (1) 1.5.2 Explain how the source works. (1) 1.6 What is the main source of energy for the Earth? 1.6.1 Name the source. (1) 1.6.2 Explain how the source works. (1)	
Section B 1.7 The equation is: _____ 1.8 What would be the effect of increasing the concentration of the reactants? 1.9 The rate of reaction "increases" when: 1.10 When is the system "equilibrium"? 1.11 What is the effect of the following on a system? 1.12 The energy in a living system is used for: 1.13 The rate of decomposition of hydrogen peroxide is affected by: 1.14 The rate of decomposition of hydrogen peroxide is affected by:	1.7 The equation is: _____ 1.8 What would be the effect of increasing the concentration of the reactants? 1.9 The rate of reaction "increases" when: 1.10 When is the system "equilibrium"? 1.11 What is the effect of the following on a system? 1.12 The energy in a living system is used for: 1.13 The rate of decomposition of hydrogen peroxide is affected by: 1.14 The rate of decomposition of hydrogen peroxide is affected by:	



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