

Assessment 3

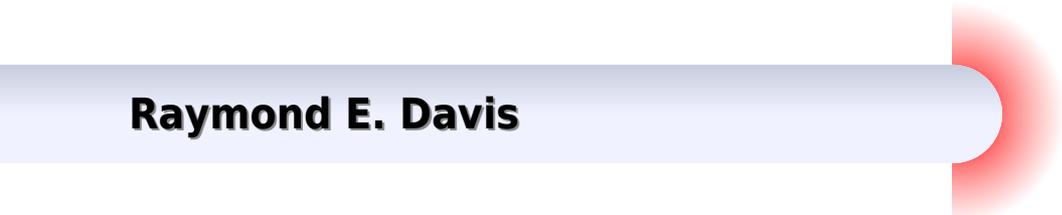
Chapter Test B**Chapter: Chemical Equations and Reactions**

Part 1 In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The production of a slightly soluble solid compound by a double displacement reaction is called the formation of a
- gas.
 - precipitate.
 - redox reaction.
 - hydrogen.
- _____ 2. To balance a chemical equation, it is permissible to adjust the
- coefficients.
 - subscripts.
 - formula of the products.
 - number of products.
- _____ 3. In a chemical equation, the symbol (aq) indicates that the substance is
- solid.
 - dissolved in water.
 - an acid.
 - molten.
- _____ 4. The tendency for a replacement reaction to occur depends on the
- activity between any two elements in the activity series themselves.
 - temperature difference.
 - valence electrons and mass of.
 - activity between any two elements in the activity series themselves.
- _____ 5. The coefficients in a chemical equation
- indicate the number of moles of each substance that react.
 - show the number of grams of each substance that react.
 - are the molar masses of the substances.
 - show the valence electrons for each atom.
- _____ 6. If metal X is more reactive than metal Y in the activity series, then
- X will replace ions of Y in solution.
 - Y will replace ions of X in solution.
 - Y will form oxides only indirectly.
 - X will react with cold water.

Holt Chemistry Answer Key Chap 8

Raymond E. Davis



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Holt Chemistry Ralph Thomas Myers,2004 **Holt Chemistry File** ,1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket **Books in Print Supplement** ,2002 Essentials of Modern Chemistry

Charles Elwood Dull,1918 **Instructor's Guide and Answer Keys to Accompany The Holt Handbook** ,2002

Introduction to Modern Chemistry Edward Florian Neuzil,1968 **Modern Chemistry** Raymond E. Davis,1999 2000
2005 State Textbook Adoption Rowan Salisbury **Introduction to Modern Chemistry: Student Manual** Eugene Meyer,1979

Children's Books in Print, 2007 ,2006 Lab Experiments Modern Chemistry Harold Clark

Metcalf,1982 Journal Royal Society for the Encouragement of Arts, Manufactures and Commerce,1868 **Study and Problem Solving Guide to Accompany Principles of Modern Chemistry, Oxtoby/Nachtrieb** Wade A. Freeman,1986

JOURNAL OF THE SOCIETY OF ARTS AND OF THE INSTITUTIONS IN UNION ,1868 **Journal of the Royal Society of Arts** Royal Society for the Encouragement of Arts, Manufactures and Commerce,1868 **Modern Chemistry** H. Clark

Metcalf,Joseph F. Castka,1966 **Principles of Modern Chemistry** David W. Oxtoby,Norman H. Nachtrieb,1996

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