

Appendix E

SELECTED THERMODYNAMIC DATA AT 298.15 K

Species	ΔH_f° (kJ mol ⁻¹)	S° (J mol ⁻¹ K ⁻¹)	ΔG_f° (kJ mol ⁻¹)	Species	ΔH_f° (kJ mol ⁻¹)	S° (J mol ⁻¹ K ⁻¹)	ΔG_f° (kJ mol ⁻¹)
Aluminum				Bismuth			
Al(s)	0	28.3	0	Bi(s)	0	56.74	0
AlCl ₃ (s)	-704.2	110.7	-628.9	BiCl ₃ (s)	-379.1	177.0	-315.0
Al ₂ O ₃ (s)	-1676	50.92	-1582	Bi ₂ O ₃ (s)	-573.88	151.5	-493.7
AlPO ₄ (s)	-1733.8	90.76	-1617.9	Bi ₂ S ₃ (s)	-143.1	200.4	-140.6
Al ₂ (SO ₄) ₃ (s)	-3440.84	239.3	-3099.94	Boron			
Antimony				B(s)	0	5.86	0
Sb(s)	0	45.69	0	BCl ₃ (l)	-427.2	206.3	-387.4
Sb ₂ O ₃ (s)	-1417.1	246.0	-1253.0	B ₂ H ₆ (g)	35.6	232.11	86.7
Sb ₂ O ₅ (s)	-971.9	125.1	-829.2	BF ₃ (g)	-1137.00	254.12	-1120.33
SbCl ₃ (l)	-440.2	301	-350.1	H ₃ BO ₃ (s)	-1094.23	88.83	-968.92
SbCl ₅ (s)	-382.17	184.1	-259.4	NaBH ₄ (s)	-183.34	104.68	-119.54
Argon				Bromine			
Ar(g)	0	154.843	0	Br(g)	111.8	174.9	82.4
Arsenic				Br ₂ (l)	0	152.23	0
As(s)	0	35.1	0	Br ₂ (g)	30.91	245.4	3.14
H ₃ As(g)	66.44	222.78	68.93	BrF ₃ (g)	-255.6	292.4	-229.5
As ₂ O ₃ (s)	-924.87	105.4	-782.3	HBr(g)	-36.4	198.59	-53.43
AsCl ₃ (l)	-305.0	216.3	-259.4	Cadmium			
Barium				Cd(s)	0	51.76	0
Ba(s)	0	62.8	0	CdCl ₂ (s)	-391.50	115.27	-343.93
BaCl ₂ (s)	-860.1	126	-810.9	CdO(s)	-258.2	54.8	-228.4
BaCO ₃ (s)	-1216.3	112.1	-1137.6	CdS(s)	-161.9	64.9	-156.5
BaF ₂ (s)	-1207.1	96.36	-1156.8	Calcium			
Ba(NO ₃) ₂ (s)	-922.07	213.8	-796.59	Ca(s)	0	41.6	0
BaO(s)	-553.5	70.42	-525.1	Ca(g)	192.6	154.8	158.9
BaSO ₄ (s)	-1465	132	-1353	Ca ²⁺ (aq)	-542.8	-53.1	-553.5
Beryllium				CaBr ₂ (s)	682.8	130	-663.6
Be(s)	0	9.54	0	CaC ₂ (s)	-62.8	70.3	-67.8
BeCl ₂ (s)	-490.4	82.68	-445.6	CaCO ₃ (s)	-1207	92.9	-1129
BeF ₂ (s)	-1026.8	53.35	-979.4	CaCl ₂ (s)	-795.0	114	-750.2
BeO(s)	-609.6	14.14	-580.3	CaF ₂ (s)	-1215	68.87	-1162
Be(OH) ₂ (s)	-907.1	50.2	-817.6	CaH ₂ (s)	-189	42	-150
BeSO ₄ (s)	-1205.20	77.91	-1093.80	CaO(s)	-635.5	40	-604.2

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Ralph Thomas Myers



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Study and Problem Solving Guide to Accompany Principles of Modern Chemistry, Oxtoby/Nachtrieb Wade A. Freeman,1986 Hearings, Reports and Prints of the House Committee on Post Office and Civil Service United States. Congress. House. Committee on Post Office and Civil Service,1965 **General Chemistry** Brady,1990-01-02 **Postal Rates for Classroom Training Cards, Hearing Before a Subcommittee on Postal Rates.....89-1, on H.R. 2837 and H.R. 2838, September 29, 1965** United States. Congress. House. Post Offices and Civil Service,1965 **Hearings** United States. Congress. House. Committee on Post Office and Civil Service,1965 Scientific, Medical Reginald Robert Hawkins,1950 Chemistry Stanley R. Radel,Marjorie H. Navidi,1990 **Polish Journal of Chemistry** ,1979 **Scientific, Medical, and Technical Books Published in the United States of America, 1930-1944** Reginald Robert Hawkins,1950 School Science and Mathematics ,1924 **Chemicals** ,1927 Chemical, Color and Oil Record ,1927

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