

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

- | | | | |
|----|----|-------|--------|
| 10 | 11 | ion | octet |
| 12 | 13 | anion | cation |

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?
They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

Holt Chemistry Concept Review Phase Equilibrium Answers

Shasha Hu



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Holt Chemistry Ralph Thomas Myers,2004 *Modern Chemistry* Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2001 **Essays in Physical Chemistry** William Thomas Lippincott,1988 **Choice** ,1966 *National Library Service Cumulative Book Review Index, 1905-1974: Titles. [A-Z* National Library Service Corporation,1975 *Evolution and Mineralization of the Arabian-Nubian Shield* Ahmed M. Shanti,1979 **Holt Chemistry** ,2003 **Holt Chemistry File** ,1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket **Holt Chemistry R.** Thomas Myers,2000 **Holt Chemistry** ,1996 **Brief Review in Chemistry** Patrick Kavanah,2002 **Sterling Test Prep AP Chemistry Review: Complete Content Review** Sterling Test Prep,2017-07-25 Recommended by teachers Trusted by students Higher score money back guarantee AP Chemistry Complete Content Review provides a detailed and thorough review of topics tested on the AP Physics 1 exam in 2018 The content covers foundational principles and concepts necessary to answer related questions on the test Electronic and atomic structure of matter Periodic table Chemical bonding States of matter gases liquids solids Solution chemistry Acids and bases Stoichiometry Equilibrium and reaction rates Thermochemistry Electrochemistry This book provides a detailed and thorough review of topics tested on the AP Chemistry exam The content covers foundational principles and theories necessary to answer related questions on the test The information is presented clearly and organized in a systematic way to provide students with targeted AP Chemistry review tool You can focus on one knowledge area at a time to learn and fully comprehend important concepts and theories or to simply refresh your memory By reading these review chapters thoroughly you will learn important chemistry concepts and the relationships between them This will prepare you for the exam and you will increase your score Scoring high on AP exams is important for admission to college To achieve a high score on AP Chemistry you need to develop skills to properly apply the science knowledge you have to solving each question Understanding key concepts having the ability to extract information from the provided data and distinguishing between similar answer choices is more valuable than simply memorizing terms All the material in this book are prepared by chemistry instructors with years of experience in applied chemistry as well as in academic settings This team of experts analyzed the content of the test released by the College Board and designed essential review that will help you build and solidify the knowledge necessary for your success on the exam The content was reviewed for quality and effectiveness by our science editors who possess extensive credentials are educated in top colleges and universities and have years of teaching and editorial experience

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