

B Chemistry – 46

Topic 2 Questions

1. Which of the following salts forms coloured solutions when dissolved in water?

- I. NiCl_2
- II. FeCl_3
- III. NaCl
- IV. ZnCl_2

- A. I and II only
- B. II and III only
- C. III and IV only
- D. I, II, III and IV

(Total 1 mark)

2. Which is an essential feature of a ligand?

- A. a negative charge
- B. an odd number of electrons
- C. the presence of two or more atoms
- D. the presence of a non-bonding pair of electrons

(Total 1 mark)

3. Which equation represents the third ionisation energy of an element M?

- A. $\text{M}^+(g) \rightarrow \text{M}^{2+}(g) + e^-$
- B. $\text{M}^{2+}(g) \rightarrow \text{M}^{3+}(g) + e^-$
- C. $\text{M}(g) \rightarrow \text{M}^{2+}(g) + 2e^-$
- D. $\text{M}^{2+}(g) \rightarrow \text{M}^{3+}(g) + e^-$

(Total 1 mark)

4. Which electrons are lost by an atom of iron when it forms the Fe^{2+} ion?

- A. One s orbital electron and two d orbital electrons
- B. Two s orbital electrons and one d orbital electron
- C. Three s orbital electrons
- D. Three d orbital electrons

(Total 1 mark)

5. Which properties are typical of d-block elements?

- I. complex ion formation

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GBP Editorial



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