

Acid Base test review Sheet

- List three properties of acids: _____
- List three properties of bases: _____
- Acids turn litmus _____ and bases turn it _____
- Name the following acids: Put a check next to it if it is a strong acid
 HBr _____ H₂SO₄ _____ HClO₄ _____
 HNO₃ _____ HI _____ H₂S _____
 HNO₂ _____
- What is the formula for carbonic acid? _____ chloric acid _____ hydrochloric acid _____
- What is the formula for finding the pH of an acid? _____
- What is the [H₃O⁺] of a neutral solution? _____ what is the [OH⁻] of a neutral solution? _____
- What is the K_w of water? _____

Define :

- Arrhenius Acid _____
- Arrhenius base _____
- Bronsted-Lowry acid _____
- Bronsted-Lowry base _____
- Lewis acid _____
- Lewis base _____
- indicator _____
- titration _____

19. Fill in the following table;

[H ⁺]	pH	[OH ⁻]	pOH	Acid or base
10 ⁻¹¹				
	5.6			
		2.27 x 10 ⁻⁹		
			8.33	

20. The strongest acid would have a pH of _____, the strongest base would have a pH of _____

21. What are the products of a neutralization reaction?

22. Write the balanced equation for the neutralization of Ca(OH)₂ and HCl

General Chemistry Acid Base Review Sheet

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