

Grade 7

Natural Sciences

Term 4

Exam paper 1

Section	Questions	Answers
Section A Name 3 elements: _____ Instructions: 1. Label each circle with an element name. 2. Write an element symbol under each circle. 3. Write the atomic number of each element. 4. Write the number of protons and neutrons in each nucleus. 5. Write the number of electrons in each atom. 6. Write the number of valence electrons in each atom. 	QUESTION 1 Diagram 1 shows a neutral atom with 11 protons and 11 electrons. The diagram is divided into three regions: A, B and C. 1.1 What is the best description for the atom? <input type="radio"/> A) It is a neutral atom. <input type="radio"/> B) It is a cation. <input type="radio"/> C) It is an anion. <input type="radio"/> D) It is a molecule. 1.2 What is the best description for the nucleus? <input type="radio"/> A) It is a neutral atom. <input type="radio"/> B) It is a cation. <input type="radio"/> C) It is an anion. <input type="radio"/> D) It is a molecule. 1.3 Why do the electron shells attract the nucleus? <input type="radio"/> A) To increase the weight. <input type="radio"/> B) To increase the number of protons. <input type="radio"/> C) To increase the number of neutrons. <input type="radio"/> D) To increase the number of electrons. 1.4 Which type of energy does the atom have? <input type="radio"/> A) Kinetic energy. <input type="radio"/> B) Potential energy. <input type="radio"/> C) Chemical energy. <input type="radio"/> D) Nuclear energy. 1.5 Which type of energy does the atom have? <input type="radio"/> A) Kinetic energy. <input type="radio"/> B) Potential energy. <input type="radio"/> C) Chemical energy. <input type="radio"/> D) Nuclear energy.	ANSWERS 1.1 D 1.2 A 1.3 C 1.4 B 1.5 B
Section B 1.6 The diagram shows a cell. The cell is divided into two parts: A and B. 1.6.1 What is the best description for the cell? <input type="radio"/> A) It is a neutral atom. <input type="radio"/> B) It is a cation. <input type="radio"/> C) It is an anion. <input type="radio"/> D) It is a molecule. 1.6.2 What is the best description for the nucleus? <input type="radio"/> A) It is a neutral atom. <input type="radio"/> B) It is a cation. <input type="radio"/> C) It is an anion. <input type="radio"/> D) It is a molecule. 1.6.3 Why do the electron shells attract the nucleus? <input type="radio"/> A) To increase the weight. <input type="radio"/> B) To increase the number of protons. <input type="radio"/> C) To increase the number of neutrons. <input type="radio"/> D) To increase the number of electrons. 1.6.4 Which type of energy does the atom have? <input type="radio"/> A) Kinetic energy. <input type="radio"/> B) Potential energy. <input type="radio"/> C) Chemical energy. <input type="radio"/> D) Nuclear energy. 1.6.5 Which type of energy does the atom have? <input type="radio"/> A) Kinetic energy. <input type="radio"/> B) Potential energy. <input type="radio"/> C) Chemical energy. <input type="radio"/> D) Nuclear energy.	ANSWERS 1.6.1 D 1.6.2 A 1.6.3 C 1.6.4 B 1.6.5 B	



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