

ICE Practice Problems

#1 – Relatively easy, no ICE table required because eq'm concentrations are given

For the reaction $\text{CH}_4(\text{g}) + \text{H}_2\text{O}(\text{g}) \leftrightarrow \text{CO}(\text{g}) + 3\text{H}_2(\text{g})$ @ 1500°C an equilibrium mixture of these gases was found to have the following concentrations $[\text{CO}] = 0.300\text{M}$, $[\text{H}_2] = 0.800\text{M}$ and $[\text{CH}_4] = 0.400\text{M}$. K_c @ $1500^\circ\text{C} = 5.67$. Determine the equilibrium concentration of H_2O in this mixture.

#2 – Requires an ICE table because you do not know the equilibrium concentrations- no product is yet formed

For the reaction $\text{CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \leftrightarrow \text{CO}(\text{g}) + \text{H}_2(\text{g})$ calculate the equilibrium concentrations of all species if 1.000 mol of each reactant is mixed in a 1.000L flask. $K_c = 5.10$ at the temperature of this reaction.

		$\text{CO}_2(\text{g})$	+	$\text{H}_2\text{O}(\text{g})$	\leftrightarrow	$\text{CO}(\text{g})$	+	$\text{H}_2(\text{g})$
[Initial]	I	1.000		1.000		0		0
[Change in]	C							
[Equilibrium]	E							

#3 – Requires an ICE table because you do not know the equilibrium concentrations. The initial concentrations must be calculated- no product is yet formed

For the reaction $\text{H}_2(\text{g}) + \text{F}_2(\text{g}) \leftrightarrow 2\text{HF}(\text{g})$ calculate the equilibrium concentrations of all species if 3.000 mol of each reactant was added 1.500L flask. K_c at the temperature of the reaction is 1.15×10^2 .

		$\text{H}_2(\text{g})$	+	$\text{F}_2(\text{g})$	\leftrightarrow	$2\text{HF}(\text{g})$
[Initial]	I	2.00		2.00		0
[Change in]	C					
[Equilibrium]	E					

#4 – Requires an ICE table because you do not know the equilibrium concentrations. Initial concentrations of reactants are given.

0.200mol of H_2 , 0.200mol of I_2 , and 0.200mol of HI were placed in a 1.00 L flask and allowed to come to equilibrium. The K_c value of the reaction at this temperature is 49.5. Determine the equilibrium concentrations of all species.

		$\text{H}_2(\text{g})$	+	$\text{I}_2(\text{g})$	\leftrightarrow	$2\text{HI}(\text{g})$
[Initial]	I					
[Change in]	C					
[Equilibrium]	E					

#5 – Requires an ICE table because you do not know the equilibrium concentrations. Initial concentrations of reactants must be calculated and no product is yet formed

For the reaction $\text{H}_2(\text{g}) + \text{F}_2(\text{g}) \leftrightarrow 2\text{HF}(\text{g})$ calculate the equilibrium concentrations of each species if 3.000 mol of H_2 and 6.000mol of F_2 are mixed in a 3.000L flask. K_c at this temperature is 1.15×10^2 .

		$\text{H}_2(\text{g})$	+	$\text{F}_2(\text{g})$	\leftrightarrow	$2\text{HF}(\text{g})$
[Initial]	I					
[Change in]	C					
[Equilibrium]	E					

Ice Practice Problems Answer Key

Prentice-Hall Staff



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Ice Practice Problems Answer Key: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous captivating novels captivating the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the engaging narratives that have captivated audiences this year. The Must-Read : Colleen Hoovers "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These bestselling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of compelling stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and gripping novel that will keep you speculating until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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