

# Chemistry Notes - Grade 10

## 5.1 Properties and Changes

**Physical Change:** Water to Ice

**Chemical Change:** Hydrogen peroxide to Water

**Physical Property:**

- ↳ Characteristic/description of the substance
- ↳ Based on Qualitative Observations (senses)
- ↳ Also based on Quantitative Observations (calculations and measurements) (e.g. colour, texture, density, smell)

**Chemical Property:**

- ↳ Describes how a substance behaves in the presence of another substance (e.g. how it reacts with other elements and compounds)
- ↳ Examples are flammability, corrosion, bleaching, rusting etc.

**Physical Change:**

- ↳ No new substance created
- ↳ Most can be reversed
- ↳ E.g. Change of state, dissolving sugar in water, cutting paper

**Chemical Change:**

- ↳ Produces a new substance
- ↳ Most can't be reversed
- ↳ E.g. Forest fire, boiling an egg, burning wood

*Five Clues that a Chemical Change has Occurred:*

1. A new colour is observed
2. A new odour is produced
3. Heat or light is produced or absorbed
4. Bubbles of gas are formed
5. A solid material (precipitate) forms in a liquid

## 5.4 Patterns in the Periodic Table

**Where are the metals, non metals, and metalloids located on the periodic table?**

- ↳ *Metals:* Left side
- ↳ *Metalloid:* On the staircase
- ↳ *Non Metals:* Right side exception being Hydrogen

**Where are the alkali metals found? What are their properties?**

- ↳ *Location:* Group 1
- ↳ *Properties:* Highly Reactive, shiny, soft, low melting point, and they have one valence electrons, soft, ductile

# Grade Chemistry Notes

**L Reisser**



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