

SS2 CHEMISTRY FIRST TERM SCHEME OF WORK

1. PERIODIC TABLE [GROUPS AND PERIODS], PROPERTIES
2. PERIODIC TABLE [FAMILIES OF ELEMENTS]
3. MASS AND VOLUME RELATIONSHIPS IN REACTIONS [ACIDS AND BASES TITRATION PRACTICAL]
4. MASS AND VOLUME RELATIONSHIPS IN REACTIONS
5. CONTINUOUS ASSESSMENT
6. VOLUMETRIC AND QUALITATIVE ANALYSIS – PRACTICAL
7. MID-TERM BREAK
8. TYPES OF CHEMICAL REACTIONS
9. OXIDATION AND REDUCTION
10. OXIDATION AND REDUCTION – REDOX REACTION
11. ELECTROLYSIS
12. REVISION
13. EXAMINATION.

THE PERIODIC TABLE

Periodic means recur at regular intervals. When elements are systematically arranged in order of ascending atomic number, certain characteristics recur at regular intervals.

Dmitri Ivanovich Mendeleev, a Russian chemist, published the first version of the periodic table in 1869. With this arrangement, certain elements with similar chemical properties and electronic structures appear in a given group.

The periodic law states that the properties of the elements are the periodic function of their atomic number.

Periodic table

The elements that appear in a vertical column are said to belong to the same group or family. They have the same number of outer electrons or valence electrons and therefore exhibit similar chemical behaviour. The elements in the same horizontal row are said to belong to the same period.

Period- Elements in the same row are in the same period and have the same number of shells.

The elements in group I are Lithium (Li), sodium (Na), potassium (K), rubidium (Rb), Caesium (Cs) and Francium (Fr). These are known as the alkali metals.

The elements in group II are called the alkaline earth metals. The alkaline earth metals are beryllium

First Term Scheme Of Work For Jssphe

Huangqi Zhang



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Table of Contents First Term Scheme Of Work For Jssphe

1. Understanding the eBook First Term Scheme Of Work For Jssphe
 - The Rise of Digital Reading First Term Scheme Of Work For Jssphe
 - Advantages of eBooks Over Traditional Books
2. Identifying First Term Scheme Of Work For Jssphe
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an First Term Scheme Of Work For Jssphe
 - User-Friendly Interface
4. Exploring eBook Recommendations from First Term Scheme Of Work For Jssphe
 - Personalized Recommendations
 - First Term Scheme Of Work For Jssphe User Reviews and Ratings
 - First Term Scheme Of Work For Jssphe and Bestseller Lists
5. Accessing First Term Scheme Of Work For Jssphe Free and Paid eBooks
 - First Term Scheme Of Work For Jssphe Public Domain eBooks
 - First Term Scheme Of Work For Jssphe eBook Subscription Services
 - First Term Scheme Of Work For Jssphe Budget-Friendly Options

6. Navigating First Term Scheme Of Work For Jssphe eBook Formats
 - ePub, PDF, MOBI, and More
 - First Term Scheme Of Work For Jssphe Compatibility with Devices
 - First Term Scheme Of Work For Jssphe Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of First Term Scheme Of Work For Jssphe
 - Highlighting and Note-Taking First Term Scheme Of Work For Jssphe
 - Interactive Elements First Term Scheme Of Work For Jssphe
8. Staying Engaged with First Term Scheme Of Work For Jssphe
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers First Term Scheme Of Work For Jssphe
9. Balancing eBooks and Physical Books First Term Scheme Of Work For Jssphe
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection First Term Scheme Of Work For Jssphe
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine First Term Scheme Of Work For Jssphe
 - Setting Reading Goals First Term Scheme Of Work For Jssphe
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of First Term Scheme Of Work For Jssphe
 - Fact-Checking eBook Content of First Term Scheme Of Work For Jssphe
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

-
- Interactive and Gamified eBooks

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