

# Class 11

# Chemistry Book

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## STOICHIOMETRY Chapter

LEARNING OBJECTIVES

Students will be able to:

- Describe atomic and molecular masses with examples.
- Determine Avogadro's number and describe the relationship between mass and Avogadro's number.
- Derive molarity of ions, empirical formulae and find their applications in solving numericals.
- Perform stoichiometric calculations with limiting reagent using molar, empirical formulae, percent and volumetric composition.
- Identify the limiting reagent in a reaction.
- Show the limiting reagent in a reaction and calculate the maximum amount of products produced and the amount of any excess reagent.
- Calculate theoretical yield, actual yield, percent yield and reagent excess.
- Calculate theoretical cell EMF and the standard cell potential.

### EXERCISES

The world is full of examples of chemical reactions. Can you imagine how world is changing? The amount of products formed when some known quantities of reactants react is a chemical change. Think about an automotive engineer studying the performance of various engines. How much air (oxygen) will be produced due to the combustion of gasoline in an automotive engine? Or suppose you are in your kitchen to make 12 cups of coffee and you start with adequate quantity of sugar and milk. How much do calculate the amount of milk required to be added? You need to be able to compare the amount in other questions without the knowledge of stoichiometry.

# First Year Chemistry Sindh Text Board

**ML Morrison**



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