

## General Chemistry I Laboratory Report

Name: \_\_\_\_\_

Lab day: \_\_\_\_\_

Date of experiment: \_\_\_\_\_

TA: \_\_\_\_\_

**10. Redox reactions and Electrochemical cells**Purposes of the experiment and brief overview of experimental method(s)Results and Observations:Part I qualitative observations

For the tests you carried out, write the observations you made, and write balanced net ionic reactions below. If nothing happened in a test, write 'no reaction'.

Reactants	Observations
$Zn + Cu^{2+}$	Zinc turns black. $Zn + Cu^{2+} \rightarrow Cu + Zn^{2+}$
$Zn^{2+} + Cu$	No reaction. $Cu + Zn^{2+} \rightarrow Zn + Cu^{2+}$
$Zn^{2+} + Pb$	No reaction. $Pb + Zn^{2+} \rightarrow Zn + Pb^{2+}$
$Zn + Pb^{2+}$	Zinc turns black. $Zn + Pb^{2+} \rightarrow Pb + Zn^{2+}$
$Pb^{2+} + Cu$	No reaction. $Cu + Pb^{2+} \rightarrow Pb + Cu^{2+}$
$Pb + Cu^{2+}$	Lead metal turns black. $Pb + Cu^{2+} \rightarrow Zn + Cu^{2+}$
$Zn + HCl$	Zinc reacts with hydrochloric acid to form bubbles.

# General Chemistry Lab Manual Answers Redox Lab

**R Bogdan**



**General Chemistry Lab Manual Answers Redox Lab:**

## **General Chemistry Lab Manual Answers Redox Lab** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**General Chemistry Lab Manual Answers Redox Lab**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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