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# ACS Gen Chem 1 Exam

**Ion** - A species of an element in which the number of electrons does not equal the number of protons.

**Isotope** - Each isotope of an atom has a different number of neutrons.

**Isoelectric** - Two atoms with the same charge.

**Isotopic** - Two atoms with the same number of neutrons.

**Transition Metal Orbitals** - The 4s orbital is filled before the 3d orbital, but listed after.

**Paramagnetic** - Atoms that possess a permanent magnetic charge due to the presence of unpaired electrons.

**Electropositive** - Tending to lose electrons and form positive ions.

**Tetrahedral Bond Angle** -  $109.5^\circ$

**Resonance Structures** - Two equally correct arrangements of one molecule's electrons; generally the molecule exists as an average of the two.

**Polar Molecule** - A molecule that is mostly positive on one side and mostly negative on the other.

**$sp^3$  Orbital** - An orbital composed of one s orbital and three p orbitals; can hold eight total electrons. Tetrahedral.

**Trigonal Planar** - The way that three objects will arrange themselves around a central atom. Non-polar.

**Free Radicals** - Compounds with an odd number of electrons in their Lewis Structure. Eg.  $\text{NO}_2$

**Incomplete Octets** - Occur in elements that can form stable bonds with less than eight electrons. Eg.  $\text{BF}_3$

**Expanded Octets** - Occur in the third row of the periodic table and beyond in elements that can form stable bonds with more than eight electrons. Eg.  $\text{XeF}_2$

**Hydrogen Bonding** - When hydrogen is attracted to electronegative atoms.

# Gen Chem Study Guide For Acs Exam

**R Sandford**



## **Gen Chem Study Guide For Acs Exam:**

## The Enigmatic Realm of **Gen Chem Study Guide For Acs Exam**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Gen Chem Study Guide For Acs Exam** a literary masterpiece penned by a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those that partake in its reading experience.

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