

[4225] - 102

M.Sc.

**INORGANIC CHEMISTRY**  
**CH - 130 : Inorganic Chemistry - I**  
**(2008 Pattern) (Semester - I) (Part - I)**

Time : 3 Hours]

[Max. Marks : 80

Instructions to the candidates :

- 1) All questions are compulsory.
- 2) Answers to the two sections should be written in separate answer books.
- 3) Neat and labelled diagrams must be drawn whenever necessary.
- 4) Figures to the right indicate full marks.

SECTION - I

**Q1)** Attempt any three of the following : **[15]**

- a) Sketch and explain the following symmetry elements in  $[B_4F_4]^-$  ion and classify it into appropriate point group.
  - i) Plane of symmetry.
  - ii) Centre of inversion.
- b) Using great orthogonality theorem, derive the character table for  $D_2$  point group. Given the classes of  $D_2$  point group are,  $E$ ,  $C_2^z$ ,  $C_2^y$ , and  $C_2^x$ .
- c) Develop stereographic projections for the following molecules. Justify your answer.
  - i) Staggered ferrocene.
  - ii) Cyclopentadienyl anion.
- d) By schematic representations, give the products of following symmetry operations in  $NF_3$  molecule.
  - i)  $C_3 \times C_3^{-1}$ .
  - ii)  $C_3 \times \sigma_{V_1}$ .
  - iii)  $\sigma_{V_1} \times \sigma_{V_2}$ .

**Q2)** Attempt any three of the following : **[15]**

- a) For a  $X_cF_4$  molecule, find out reducible representation for which sigma bonds form the basis and find out which of the orbitals from the  $X_c$  atom will be offered for sigma bonding.  
Given : Character table for  $D_{4h}$  point group.
- b) Classify any two of the following molecules into appropriate point group. Justify it.
  - i)  $CO_3^{2-}$  (Planar).
  - ii)  $S_nF_4$ .
  - iii)  $SO_2$ .

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**CH Cherryholmes**



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