

Lab Report: Limiting Reactant and Percent Yield

The purpose of this lab is to explore the yields of a chemical reaction by identifying the limiting reactant, theoretical, percent, and experimental yields, while determining sources of error that may occur throughout the lab.

Throughout the lab, the question being focused on is "How can you determine the limiting reactant during a chemical reaction?"

The hypothesis is "If a substance is the limiting reactant, then it will be fully consumed by the time the reaction is complete because it is the reactant that is used up first"

- The independent variable in this experiment is the amount of aluminum used in the reaction. Since in trial 1 it was 0.50g, while in trial 2 it was 0.25g.
- The dependent variable is which reactant is the limiting variable since it changed depending on the quantity of aluminum.
- Some constants are the amount of copper(II) chloride used, 2.50g, and the equipment used.

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