

Name: _____ Date: _____

Student Exploration: Limiting Reactants

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Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript.

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

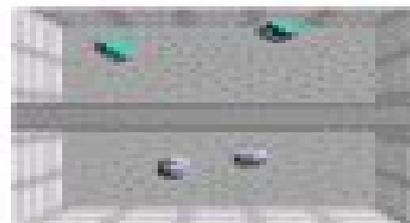
Imagine you and your friends are making hot dogs. A complete hot dog consists of a roller and a bun. At the store, you buy four packages of eight rollers and three bags of 10 buns.

1. How many total hot dogs can you make? _____
2. Which ingredient limited the number of hot dogs you could make? _____
3. Which ingredient will you have leftovers of? _____

Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the Limiting Reactants Gizmo™, you can determine which reactant is limiting in various scenarios.



To begin, make sure $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ is selected. The small "2" in H_2 , O_2 , and H_2O is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

1. Use the sliders to set the number of O_2 molecules and H_2 molecules to two.
 - A. How many hydrogen molecules (H_2) are there? _____ Hydrogen atoms? _____
 - B. How many oxygen molecules (O_2) are there? _____ Oxygen atoms? _____
2. How many H_2O molecules do you think will form when these four molecules react? _____
3. Click **Play** (▶). How many H_2O molecules actually formed? _____

Limiting Reactants Answer Key

J Dewey



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Table of Contents Limiting Reactants Answer Key

1. Understanding the eBook Limiting Reactants Answer Key
 - The Rise of Digital Reading Limiting Reactants Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Limiting Reactants Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Limiting Reactants Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Limiting Reactants Answer Key
 - Personalized Recommendations
 - Limiting Reactants Answer Key User Reviews and Ratings
 - Limiting Reactants Answer Key and Bestseller Lists

5. Accessing Limiting Reactants Answer Key Free and Paid eBooks
 - Limiting Reactants Answer Key Public Domain eBooks
 - Limiting Reactants Answer Key eBook Subscription Services
 - Limiting Reactants Answer Key Budget-Friendly Options
6. Navigating Limiting Reactants Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Limiting Reactants Answer Key Compatibility with Devices
 - Limiting Reactants Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Limiting Reactants Answer Key
 - Highlighting and Note-Taking Limiting Reactants Answer Key
 - Interactive Elements Limiting Reactants Answer Key
8. Staying Engaged with Limiting Reactants Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Limiting Reactants Answer Key
9. Balancing eBooks and Physical Books Limiting Reactants Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Limiting Reactants Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Limiting Reactants Answer Key
 - Setting Reading Goals Limiting Reactants Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Limiting Reactants Answer Key
 - Fact-Checking eBook Content of Limiting Reactants Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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