

Name : _____

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Limiting Reactant Worksheet

- 1 Balance the following reaction:



- a) Suppose 100 grams of sodium chloride reacts with 200 g of barium phosphate. What is the limiting reactant?

- b) How many grams of excess reactant are left?

- 2 Acrylonitrile ($\text{C}_3\text{H}_3\text{N}$) is the starting material for the production of a kind of synthetic fiber acrylic. It can be made from propylene (C_3H_6) by reacting with nitric oxide (NO) as follows:



What mass of $\text{C}_3\text{H}_3\text{N}$ can be made when 21.6 g of propylene react with 21.6 g of nitric oxide?

Limiting Reactants Key

**John C. Kotz, Paul Treichel, Patrick A.
Harman**



Limiting Reactants Key:

Chemistry John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant

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Chemical Equilibria Harry L. Pardue, 2018-11-15 Concepts procedures and programs described in this book make it possible for readers to solve both simple and complex equilibria problems quickly and easily and to visualize results in both numerical and graphical forms They allow the user to calculate concentrations of reactants and products for both simple and complicated situations The user can spend less time doing calculations and more time thinking about what the results mean in terms of a larger problem in which she or he may be interested

Holt Chemistry

Ralph Thomas Myers,2004 *Comprehensive Dictionary of Chemical Engineering* Muhammad Rashid Usman,2015-03-29 This book is a comprehensive collection of chemical engineering terms in a single volume It covers generally all the chemical engineering literature and has distinguished features The book is a useful reference material for the people both at the schools and the industry The author s experience of teaching and research over the years has realized a must book of this kind The terms are written in alphabetical order Where a term deserves more elaboration a rather detailed description is provided The book also contains a number of labeled diagrams which may be helpful in understanding some critical terms

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Essential Concepts of Chemistry Sharon Sherman, Alan Sherman, 1999 Designed especially for students who have little or no background in chemistry or mathematics Essential Concepts of Chemistry makes complex concepts understandable This text provides an inexpensive one color alternative for introductory chemistry courses and emphasizes everyday applications of chemistry

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