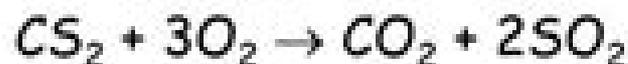


# Calculations need to be based on the limiting reactant.

- Example 2: What is the mass of sulfur dioxide that can be produced by the reaction of 95.6 g carbon disulfide with 100. g oxygen?



$$\frac{95.6 \text{ g } CS_2}{76.2 \text{ g } CS_2} \times \frac{1 \text{ mol } CS_2}{1 \text{ mol } CS_2} \times \frac{2 \text{ mol } SO_2}{1 \text{ mol } CS_2} \times \frac{64.1 \text{ g } SO_2}{1 \text{ mol } SO_2} = 161 \text{ g } SO_2$$

$$\frac{100. \text{ g } O_2}{32.0 \text{ g } O_2} \times \frac{1 \text{ mol } O_2}{3 \text{ mol } O_2} \times \frac{2 \text{ mol } SO_2}{3 \text{ mol } O_2} \times \frac{64.1 \text{ g } SO_2}{1 \text{ mol } SO_2} = 134 \text{ g } SO_2$$

134 g of  $SO_2$  can be produced in this reaction.

# Limiting Reactant Pogil Lab

**LL Leslie**



## **Limiting Reactant Pogil Lab:**

## Reviewing **Limiting Reactant Pogil Lab**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Limiting Reactant Pogil Lab**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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