



LIMITING REAGENT AND PERCENT YIELD

Section Review

Objectives

1. Calculate limiting reagent and percent yield in a chemical reaction.
2. Calculate limiting reagent and percent yield in a chemical reaction.

Vocabulary

- | | |
|----------------------|----------------------|
| 1. limiting reagent | 2. limiting yield |
| 3. percent yield | 4. theoretical yield |
| 5. theoretical yield | |

Key Equations

1. percent yield = $\frac{\text{actual yield}}{\text{theoretical yield}} \times 100$

Part A: Completion

Use the appropriate number of significant figures to complete the following problems. Round answers to the correct number of significant figures. Write answers in the spaces provided.

1. A reaction produces 12.5 g of product. The theoretical yield is 15.0 g. What is the percent yield? _____
2. A reaction produces 10.0 g of product. The theoretical yield is 12.5 g. What is the percent yield? _____
3. A reaction produces 15.0 g of product. The theoretical yield is 10.0 g. What is the percent yield? _____
4. A reaction produces 12.5 g of product. The theoretical yield is 15.0 g. What is the percent yield? _____
5. A reaction produces 10.0 g of product. The theoretical yield is 12.5 g. What is the percent yield? _____
6. A reaction produces 15.0 g of product. The theoretical yield is 10.0 g. What is the percent yield? _____

7. _____
8. _____
9. _____
10. _____
11. _____
12. _____

Part B: Short-Answer

Identify each limiting reagent and calculate the percent yield for the following reactions.

1. _____
2. _____
3. _____

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