

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

Department:
Basic Education
REPUBLIC OF SOUTH AFRICA

G.12

NATIONAL
SENIOR CERTIFICATE

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



4.5 Write down the TYPE of absorption in reaction II.

4.6 Compounds R and T are positional isomers.

The inorganic reagents shown below are available for reactions IV and V.

Br ₂	K ₂ Cr ₂ O ₇ (aq)	K ₂ Cr ₂ O ₇ (s)	HBr	K ₂
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4.7 Define for both chemical equilibrium.

Use the graph to answer the questions below.

Grade physics 2014 Question Paper 2

LP Steffe



Grade physics 2014 Question Paper 2:

Unveiling the Magic of Words: A Overview of "**Gradephysics2014 Question Paper2**"

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