

 FIELD COMPLETE

# The Comprehensive Guide to HVAC Licensing in Texas



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# Hvac License Study Guide

**Mike Yu, Examreview**



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 $\text{GaBr}_3 + \text{NaBr} \rightarrow \text{SnO}_4$   
 $4\text{NaSO}_3 + \text{O}_2 \rightarrow 2\text{Na}_2\text{SO}_4$   
 $3\text{NF}_3 + 5\text{SnF}_2 \rightarrow \text{N}_2\text{O}_3$  using the equation from problem 2 above answer the following

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 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$   
 $2\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$   
 $3\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$   
 $2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{O}_3 + 3\text{SO}_3 + 6\text{NaBr}$   
 $3\text{SnO}_2 + 2\text{NF}_3 + 3\text{SnF}_2 \rightarrow \text{N}_2\text{O}_3$  using the following equation  
 $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 $\text{Ce} + \text{Zn} + \text{CuCl}_2 \rightarrow \text{ZnCl}_2 + \text{Cu} + \text{Zn} + \text{CuCl}_2$   
 $2\text{ZnCl}_2 + 2\text{Cu}$  how many moles of  $\text{Ce} + \text{ZnCl}_2 + \text{ZnCl}_2$  will be produced from 23.0 g of  $\text{Ce} + \text{Zn}$  assuming  $\text{Ce} + \text{CuCl}_2 + \text{CuCl}_2$  is available in excess

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 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$   
 $2\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$   
 $3\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$   
 $2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{O}_3 + 3\text{SO}_3 + 6\text{NaBr}$   
 $3\text{SnO}_2 + 2\text{NF}_3 + 3\text{SnF}_2 \rightarrow \text{N}_2\text{O}_3$  using the following equation  
 $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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