

Fill in the table on the right from your periodic table.

Element	Chemical Symbol	Atomic #	Valence Electrons	Oxidation #	# of Protons	Atomic Mass	Metal or Nonmetal	Group or Luster Electrons	# of Full Electron Levels
Oxygen									
Argon									
Calcium									
Nitrogen									
Potassium									
Sulfur									
Helium									

Find the molar mass of the following compounds

Calcium oxide (CaO)

Water

Classify (ionic, covalent or polyatomic) and name the following compounds.

- CO₂ _____
- N₂O _____
- NaClO₂ _____

Draw the Lewis Dot Diagrams for

- Carbon
- Sodium

Make balanced ionic compounds from:

- Li + O _____
- Be + NO₃⁻ _____
- Na + Ar _____
- Ca + O _____

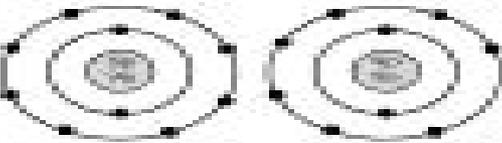
With dot diagrams draw the covalent bond of O₂

With dot diagrams the covalent bond of CH₄

1. Valence electrons are the outermost electrons of an atom that are involved in chemical bonding. True/False

2. Lithium has _____ valence electrons, will _____ (lose or gain) _____ electrons and become _____ (positive or negative). This is why lithium's oxidation number is _____.

3. A metal and non-metal will form a _____ compound, two non-metals form a _____ compound.



Which one is the ion?

Give the ion notation for the ion.

Matching:

A. Ion	If you change the number of protons, you change the _____.
B. Element	If you change the number of neutrons, you change the _____.
C. Isotope	If you change the number of electrons, you change the _____.
D. Oil	Every 3,000 miles you should change a car's _____.

Matching:

A. Unit of Measurement of Mass	_____ Unit of length; equal to 3.3 ft.
B. Matter	_____ Mass is neither destroyed or created in chemical reactions.
C. Conductor	_____ Unit of mass; about 1 dollar bill.
D. Labor	_____ 1,000's of mass; width of pinky finger.
E. Oil	_____ Unit of volume; just bigger than a quart.

Matching:

A. Solid	_____ Molecules that are tightly packed and retain their shape and size.
B. Liquid	_____ Molecules that bounce off of each other, can be compressed, and take the shape of their container.
C. Gas	_____ Molecules that can move (slide) over each other and have a definite size (volume), but not shape and can not be compressed.
D. Oil	

Matching:

A. Mixture	_____ A substance made up of two or more atoms that must be separated by chemical means.
B. Compound	_____ Something made up of 2 or more different atoms that can be separated by physical means.
C. Element	_____ A substance that is pure and made up of only one type of atom.
D. Matter	_____ Anything that has mass and takes up space.
E. Heterogeneous	_____ A mixture that is the same throughout.
F. Homogeneous	_____ A mixture that is different throughout.

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RC Schank



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