

AP[®] ENVIRONMENTAL SCIENCE
2013 SCORING GUIDELINES

Question 3 (continued)

(c) The Montreal Protocol of 1987 provided a global framework to phase out chlorofluorocarbon (CFC) production and use. Although the Montreal Protocol has led to a dramatic decrease in CFCs released into the atmosphere, stratospheric ozone destruction has decreased only slightly.

(i) Explain the process by which CFCs lead to the destruction of stratospheric ozone. (You may use chemical equations in your answer.)

(2 points: 1 point for a correct description of the decomposition of CFCs and 1 point for a correct description of the reaction of ozone with chlorine)

Correct descriptions of the decomposition of CFCs include one of the following:

- Absorption of UV radiation by CFC molecules releases chlorine atoms
- $\text{CCl}_2\text{F} + \text{UV} \rightarrow \text{CCl}_2\text{F} + \text{Cl}$

Correct descriptions of the destruction of stratospheric ozone include one of the following:

- Chlorine atoms break down ozone molecules
- $\text{Cl} + \text{O}_3 \rightarrow \text{ClO} + \text{O}_2$

(ii) Explain why the rapid decrease in CFC emissions has not led to a similarly rapid decrease in the destruction of stratospheric ozone.

(1 point can be earned for a correct explanation linking the absence of a rapid decrease in the destruction of stratospheric ozone with one of the following):

- The slow migration of CFCs into the stratosphere.
- The long lifetime of CFCs and/or chlorine in the stratosphere.
- The continued release of other ozone-depleting substances.

(d) Identify a human activity that leads to the formation of tropospheric ozone as a secondary pollutant and explain why tropospheric ozone levels peak in the daytime.

(2 points: 1 point for a correct human activity and 1 point for correctly explaining that sunlight is required to form tropospheric ozone; only the first human activity can earn points)

The following are acceptable human activities:

- Burning fossil fuels (e.g., operating motor vehicles, using landscaping equipment, electric power generation, industrial production)
- Releasing VOCs (e.g., pumping gas, using solvent-based paints)

The following are acceptable explanations for why tropospheric ozone levels peak in the daytime:

- Sunlight is required to form tropospheric ozone
- Tropospheric ozone is created by photochemical reactions

Environmental Science 2013 Response Scoring Guidelines

Marco Cascella



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Table of Contents Environmental Science 2013 Response Scoring Guidelines

1. Understanding the eBook Environmental Science 2013 Response Scoring Guidelines
 - The Rise of Digital Reading Environmental Science 2013 Response Scoring Guidelines
 - Advantages of eBooks Over Traditional Books
2. Identifying Environmental Science 2013 Response Scoring Guidelines
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Environmental Science 2013 Response Scoring Guidelines
 - User-Friendly Interface
4. Exploring eBook Recommendations from Environmental Science 2013 Response Scoring Guidelines
 - Personalized Recommendations
 - Environmental Science 2013 Response Scoring Guidelines User Reviews and Ratings
 - Environmental Science 2013 Response Scoring Guidelines and Bestseller Lists

5. Accessing Environmental Science 2013 Response Scoring Guidelines Free and Paid eBooks
 - Environmental Science 2013 Response Scoring Guidelines Public Domain eBooks
 - Environmental Science 2013 Response Scoring Guidelines eBook Subscription Services
 - Environmental Science 2013 Response Scoring Guidelines Budget-Friendly Options
6. Navigating Environmental Science 2013 Response Scoring Guidelines eBook Formats
 - ePub, PDF, MOBI, and More
 - Environmental Science 2013 Response Scoring Guidelines Compatibility with Devices
 - Environmental Science 2013 Response Scoring Guidelines Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Environmental Science 2013 Response Scoring Guidelines
 - Highlighting and Note-Taking Environmental Science 2013 Response Scoring Guidelines
 - Interactive Elements Environmental Science 2013 Response Scoring Guidelines
8. Staying Engaged with Environmental Science 2013 Response Scoring Guidelines
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Environmental Science 2013 Response Scoring Guidelines
9. Balancing eBooks and Physical Books Environmental Science 2013 Response Scoring Guidelines
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Environmental Science 2013 Response Scoring Guidelines
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Environmental Science 2013 Response Scoring Guidelines
 - Setting Reading Goals Environmental Science 2013 Response Scoring Guidelines
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Environmental Science 2013 Response Scoring Guidelines
 - Fact-Checking eBook Content of Environmental Science 2013 Response Scoring Guidelines
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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