

## ACS Gen Chem 2 Final Exam Study Guide With Correct Solutions.

C - Answer Where is the hydrophilic (attracted to water) region of the molecule?

- (A) Region 1
- (B) Region 2
- (C) Region 3
- (D) The three regions are equally hydrophilic

C - Answer Which molecule is most soluble in water?

D - Answer A solution of NaCl in water has a concentration of 20.5% by mass. What is the molal concentration of the solution?

Molar Mass NaCl = 58.44 g/mol

- (A) 0.205 m
- (B) 0.258 m
- (C) 3.51 m
- (D) 4.41 m

B - Answer What is the mole fraction of water in 200 g of 89% (by mass) ethanol, C<sub>2</sub>H<sub>5</sub>OH?

Molar Mass C<sub>2</sub>H<sub>5</sub>OH = 46 g/mol

- (A) 0.11
- (B) 0.24
- (C) 0.32
- (D) 0.76

D - Answer A mixture of 100 g of K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> and 200 g of water is stirred at 60 degrees C until no more of the salt dissolves. The resulting solution is poured off into a separate beaker, leaving the undissolved solid behind. The solution is now cooled to 20 degrees C. What mass of K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> crystallizes from the solution during the cooling?

- (A) 9 g
- (B) 18 g
- (C) 31 g
- (D) 62 g

A - Answer Carbonated beverages have a fizz because of gas dissolved in the solution. What increases the concentration of gas in a solution?

- (A) cooling the solution
- (B) heating the solution
- (C) increasing the volume of solution
- (D) decreasing the volume of solution

# Gen Chem 15final Exam Review Guide

**Tao Wei**



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## **Table of Contents Gen Chem 15final Exam Review Guide**

1. Understanding the eBook Gen Chem 15final Exam Review Guide
  - The Rise of Digital Reading Gen Chem 15final Exam Review Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Gen Chem 15final Exam Review Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Gen Chem 15final Exam Review Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Gen Chem 15final Exam Review Guide
  - Personalized Recommendations
  - Gen Chem 15final Exam Review Guide User Reviews and Ratings
  - Gen Chem 15final Exam Review Guide and Bestseller Lists
5. Accessing Gen Chem 15final Exam Review Guide Free and Paid eBooks
  - Gen Chem 15final Exam Review Guide Public Domain eBooks
  - Gen Chem 15final Exam Review Guide eBook Subscription Services
  - Gen Chem 15final Exam Review Guide Budget-Friendly Options
6. Navigating Gen Chem 15final Exam Review Guide eBook Formats

- ePub, PDF, MOBI, and More
  - Gen Chem 15final Exam Review Guide Compatibility with Devices
  - Gen Chem 15final Exam Review Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Gen Chem 15final Exam Review Guide
  - Highlighting and Note-Taking Gen Chem 15final Exam Review Guide
  - Interactive Elements Gen Chem 15final Exam Review Guide
8. Staying Engaged with Gen Chem 15final Exam Review Guide
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Gen Chem 15final Exam Review Guide
9. Balancing eBooks and Physical Books Gen Chem 15final Exam Review Guide
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Gen Chem 15final Exam Review Guide
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Gen Chem 15final Exam Review Guide
- Setting Reading Goals Gen Chem 15final Exam Review Guide
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gen Chem 15final Exam Review Guide
- Fact-Checking eBook Content of Gen Chem 15final Exam Review Guide
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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