

Lab: Mixing Around

Objectives: In previous labs, we learned that molecules are constantly in motion and that energy transfers from warmer objects to colder objects. Today we are going to figure out what effect mass has on the thermal energy of a system.

Materials:

- Electric balance
- 300mL beakers (3)
- Water
- Thermometers (3)
- Styrofoam
- Hotplates
- Stopwatch
- Hot gloves
- Tape
- Safety goggles

Procedure:

1. Place a 300mL beaker on an electric balance. Zeroing the balance for each 300mL beaker you place on the balance (DO NOT measure out the water)
2. Measure out 20g of water in the first beaker, 30g of water in the second beaker, and 40g in the third beaker.
 - a. Use the pipette to get as close as possible (within 0.1g, is acceptable)
3. Put all 3 beakers on top of the same hot plate. (DO NOT TURN IT ON YET)
4. Record the initial temperature of the water in each of the beakers.
5. Remove the thermometers and place them on a towel for later use.

General Chemistry Massing Activities Lab Answers

Thomas Griffiths



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