

Study Guide for Chemistry EOC Assessment

Your test will consist of multiple-choice questions over the following topics.

1. Metric units for mass, length, volume
2. Metric to metric conversions (K H Da $\frac{m}{l}$ /g D C M)
3. Know how to put numbers in scientific notation and take them out.
4. Significant Digit rules
5. Difference between chemical and physical properties and changes. There will be multiple questions on this.
6. Ways to tell if a chemical change (chemical reaction) is taking place.
7. Pure substances and mixtures
 - a. heterogeneous mixture
 - b. homogeneous mixture
8. Density & how it relates to floating and sinking
9. Use the equation for density
$$\text{Density} = \frac{\text{Mass}}{\text{volume}}$$

(g/ml) (g) (ml)
10. Know the phases of matter and the characteristics of each phase.
11. Law of Conservation of Matter
12. What is a mole in chemistry?
13. Convert grams to moles, and moles to grams
14. Define an atom
15. Structure of an atom and where each subatomic particle is located.
 - a. Nucleus (contains protons and neutrons)
 - b. Protons (positive)
 - c. Neutrons (neutral)
 - d. Electrons (negative)
16. Isotopes
17. Elements
18. If given atomic # and mass, find the number of protons and neutrons in an atom.
19. Find the number of valence electrons, and tell what it says about reactivity.
20. Periodic Table
 - a. Know about the families
 - b. Group vs. period
21. Know about the properties of metals and nonmetals.
 - a. Ductile, malleable, conductors
22. Ionic & Covalent Compounds
How do they form? (Transfer or gain electrons)
Characteristics of Each
23. Be able to name ionic compounds
24. Predict oxidation numbers
25. Chemical Equations:
Know the parts of an equation
5 main types of reactions-know about them and be able to identify them
(Synthesis, Decomposition, Single Displacement, Double Displacement, Combustion)
26. Acids & Bases-
Macro and Microscopic properties of each
What ions do they form, how do they affect litmus, what is their pH range?
Identify examples of each- Mono, di, and polyprotic acids
Tell if they are strong or weak
Acid-base indicator

Eoc Chemistry Study Guide

Mary Amboji



Eoc Chemistry Study Guide:

Cracking the Virginia SOL Tina Walton, Princeton Review (Firm), 2000 The Princeton Review realizes that acing the English Reading Literature and Research exam is very different from getting straight As in school The Princeton Review doesn't try to teach students everything there is to know about English only the techniques they'll need to score higher on the exam There's a big difference In **Cracking the Virginia SOL EOC English Reading Literature and Research** The Princeton Review will teach test takers how to think like the test makers and eliminate incorrect answer choices by using Process of Elimination and other techniques Familiarize students with every type of reading material that might be on the test Use the two pass system to get a grip on the test structure and raise scores Hone individual skills with practice questions for each type of test question This book includes 2 full length simulated exams All of TPR's sample test questions are just like the ones test takers will see on the actual End of Course English Reading Literature and Research exam and TPR fully explains every solution Contents Include The Mystery Test Structure and Strategies II Terms Literary Movements and Writers Literature Review III Literary Elements and Printed and Resource Materials The Elements of Literature A Variety of Printed Material A Variety of Resource Material IV The Princeton Review Practice Tests *Roadmap to the Virginia SOL* Princeton Review, 2005 *Roadmap to the Virginia SOL EOC Chemistry* includes strategies that are proven to enhance student performance The experts at The Princeton Review provide content review of the crucial material most likely to appear on the test detailed lessons complete with test taking techniques for improving test scores 2 complete practice Virginia SOL EOC Chemistry tests

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Seager/Slabaugh,2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application **Pamphlet - Dept. of the Army** United States. Department of the Army,1951 **Texas EOC Chemistry - the All-In-one Guide** Mary Amboji,2013-03-19

The complete guide for Texas EOC Chemistry has hundreds of guided and independent practice questions based on the categories tested on the EOC The dual coded questions help students gain mastery over scientific process skills Five category

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