

Scheme – I
Sample Question Paper:

Programme Name : Automobile Engineering
Programme code : AE
Semester : VI Sem
Course Title : Autotronics
Marks : 70

22654

Time: 3Hrs.

Instructions:

- (1) All questions are compulsory.
- (2) Illustrate your answers with neat sketches wherever necessary.
- (3) Figures to the right indicate full marks.
- (4) Assume suitable data if necessary.
- (5) Preferably, write the answers in sequential order.

Q.1) Attempt any FIVE of the following.

(10 Marks)

- a) Enlist two types of diodes used in Automotive Electronics.
- b) State four types of Computer Memories used in Automotive applications.
- c) State the name of the sensor used in measurement of vehicle speed.
- d) Name the two types of computer controlled fuel injection systems used in modern vehicles.
- e) Give an example of particular DTC code found in vehicle diagnosis.
- f) Draw symbol of Photo-Diode.
- g) Identify the name of a communication system used for navigation purpose in modern automobiles.

Q.2) Attempt any THREE of the following.

(12 Marks)

- a) Draw block diagram of digital visual display in vehicle instrumentation.
- b) Convert the given decimal number 100.5 into binary number.
- c) Sketch and explain signal conditioning system used for oxygen sensor.
- d) Compare open loop and closed loop control system in automotive application.

Q.3) Attempt any THREE of the following.

(12 Marks)

- a) Illustrate with neat sketch the output signal waveform for Crank shaft position sensor.
- b) Draw block diagram of GDI system.
- c) Describe stand alone diagnosis procedure of coolant temperature sensor.
- d) Construct a block diagram for Common Rail Direct Injection system used in modern vehicles.

Q.4) Attempt any Three of the following.

(12 Marks)

- a) Explain with justification the use of Bluetooth technology as a communication system in automobiles.
- b) Answer the following (Refer Figure)

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Thomas J. Hummel, Steven S. Zumdahl

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Laboratory Manual of Pharmacology Dr. Remeth J. Dias, Dr. C. R. Patil, Dr. Kuldeep U. Bansod, Dr. Kailas Mali, 2023-09-28
We are very pleased to put forth the revised edition of Laboratory Manual of Pharmacology. We have incorporated all the suggestions, modified it to make it easier, student friendly, and relevant in terms of achieving curriculum outcome. We are very much thankful to all the learned teachers who have given their feedback wholeheartedly. We have even incorporated the changes in this manual based on the feedback given by the teachers from all the institutes. Now we believe that the manual has been fulfilling the aspirations of pharmacology teachers and students too. This manual is prepared as per PCI Education Regulations 2020 for Diploma Course in Pharmacy. The methods of all the experiments are reviewed and added from the recent research papers so that the advancement in the methods or apparatus can be addressed. This manual is designed for outcome-based education, and each experiment is arranged in a uniform way such as practical significance, practical outcomes, PrOs, and its mapping with course outcomes, minimum theoretical background, resources used, procedure, precautions, observations, result, conclusion, references, and related questions. We have also given the readings for the reference of students and better understanding. Moreover, an assessment scheme is also given to help the student and teacher to know what to be assessed. The experiments given are as per the OECD guidelines. Teacher and students have to use suitable software or MSBTE software or any other software to know the demonstration of the experiment. The tables are given to record the observations from the MSBTE software or any other suitable software. In addition, the questions are given at the end of experiments to increase the knowledge of students about the application of statistics, which would be helpful for them when they will go for higher studies. Hope this manual will help the students to learn the concept, principles, and perform the experiments virtually. We wish you all the best.

Chemical Thermodynamics of Americium R.J. Silva, 1995-10-25. This is the second volume in a series of critical reviews of the chemical thermodynamic data of those elements of particular importance in the safety assessment, modeling of high-level radioactive waste storage and disposal facilities. The objective of these reviews is to provide a set of reliable thermodynamic data that can be used to describe the behaviour of these elements under conditions relevant for radioactive waste disposal systems and the geochemical environments. The present volume is a review of experimental data reported in the literature for americium. On a few occasions where no data existed, comparisons and estimates were made based on experimental data on analog lanthanide elements. The basic philosophy was to develop a minimum set of solid phases and solution species of americium that would fit all experimental data being reviewed.

Student Solution Manual for Introduction to Chemical Principles H. Stephen Stoker, Nancy J. Gardner, 2013-04-22. The Student Solutions Manual includes full solutions to all odd-numbered end-of-chapter problems in the text and answers to all multiple-choice practice test questions.

[Chemistry of Pulp and Paper Making](#) Edwin Sutermeister, 2013-09. This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book

without typos from the publisher Not indexed Not illustrated 1920 edition Excerpt Ind Eng Chem 1916 8 780 it can be caused by traces of manganese and it seems probable that it may be caused at times by either one of these substances Electrolytic Bleach Besides the various calcium hypochlorite solutions there are numerous devices for the electrolysis of salt solutions and the direct application of the sodium hypochlorite solutions thus obtained The most celebrated of these is the Hermite process which originally employed a solution of magnesium chloride but in which salt was later used almost exclusively According to this plan the solution was first electrolyzed and then passed through the material to be bleached and back to the electrolyzer thus keeping up a continuous circulation This continuous process can only be applied to rag stock as the impurities dissolved in bleaching esparto or chemical wood pulp soon contaminate the solution to such an extent as to interfere with its proper operation The Hermite process is probably not used in this country Many other schemes for the use of electrolyzed salt solutions have been proposed and much has been said about the superior bleaching power of a pound of chlorine thus prepared over the same quantity in the form of bleaching powder solution While these claims are undoubtedly made in good faith it seems probable that many are based on incorrect comparisons since tests of efficiency by bleaching for a given time and determining the residual bleach are not accurate unless exactly the same colors are produced Ahlin 1 states that it is not true that active chlorine produced electrolytically will do more work than an equal quantity from bleaching powder and Dorenfeldt² claims that unless brine is to be obtained almost free of cost or unless sodium carbonate is

Paper Chemistry J.C. Roberts, 2012-12-06 Although the title of this book is Paper Chemistry it should be considered as a text about the chemistry of the formation of paper from aqueous suspensions of fibre and other additives rather than as a book about the chemistry of the raw material itself It is the subject of what papermakers call wet end chemistry There are many other excellent texts on the chemistry of cellulose and apart from one chapter on the accessibility of cellulose the subject is not addressed here Neither does the book deal with the chemistry of pulp preparation from wood from other plant sources or from recycled fibres for there are also many excellent texts on this subject The first edition of this book was a great success and soon became established as one of the Bibles of the industry Its achievement then was to collect the considerable advances in understanding which had been made in the chemistry of papermaking in previous years and provide for the first time a sound physico chemical basis of the subject This new edition has been thoroughly updated with much new material added The formation of paper is a continuous filtration process in which cellulosic fibres are formed into a network which is then pressed and dried The important chemistry involved in this process is firstly the retention of colloidal material during filtration and secondly the modification of fibre and sheet properties so as to widen the scope for the use of paper and board products *Chemistry of Paper* J C Roberts, 2007-10-31 The manufacture of paper involves a large amount of chemistry including carbohydrate chemistry pigments and resins and colloid and surface chemistry as well as elements of environmental and analytical chemistry Providing an overview of the making of paper from a chemical

perspective this book deals with both the chemistry of paper as a material and the chemistry of its production The book explores several chemical processes involved in the production of paper the delignification of the wood fibres performed at elevated temperature and pressure the bleaching of the cellulose rich pulp using environmentally friendly systems the formation of the pulp into sheets of fibres strengthened by extensive inter fibre hydrogen bonding and finally the coating of the sheets in a manner appropriate to their end use This book is an informative and entertaining overview for students and others who require an introduction to the chemistry of paper manufacture

Complete Solutions Guide, Chemical Principles, Steven S. Zumdahl, 4th Ed Thomas J. Hummel, Steven S. Zumdahl, 2002

Chemistry of Modern Papermaking Cornel Hagiopol, JAMES W. JOHNSTON, 2021-06-30 Chemistry of Modern Papermaking presents a chemist's perspective on the papermaking process With roughly 3% of the mass of a paper product invested in water soluble chemicals paper makers can adjust the speed and efficiency of the process minimize and reuse surplus materials and differentiate a paper product as required by specific customers With research and development scattered across academic journals or the proprietary information of a variety of companies the modern paper maker requires a one stop resource for the general picture of the paper chemicals organic chemistry The remarkable effectiveness of chemicals at very low concentrations is a fascinating topic in paper chemistry Chemistry of Modern Papermaking is an extensive collection of data combining the organic chemistry of paper with its potential applications Within each chapter paper chemicals are organized based on their chemical structure After an introduction and brief history the book explores the papermaking process retention aids temporary wet strength resins wet strength resins dry strength resins internal sizing agents creping adhesives and softeners and chemicals for paper surface treatment Uniquely patents and scientific articles are included in almost equal number among the 3400 references Chemistry of Modern Papermaking focuses on the chemistry behind each application on what has been done and on what can be done Never before has a book analytically arranged and lucidly explained such an expansive collection of details from both the patent and scientific literature This synthesis is achieved not only through diligent work but also through years of industrial experience

Paper Chemistry J. C. Roberts, 1991-06 Cellulose Electrokinetics Retention aids Additives Sizing Whitening agents Fillers Measurement and control

Paper Chemistry Dan Eklund, Tom Lindström, 1991

Solved Examples in Chemical Engineering G. K. Roy, 1992

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