

Answer KeyNaming and Writing Formulas for Ionic Compounds

Write the names for the following compounds:

1. KClO_4 – potassium perchlorate
2. Sr_3P_2 – strontium phosphide
3. NH_4NO_2 – ammonium nitrite
4. NaCl – sodium chloride
5. CoS – cobalt (II) sulfide
6. ZnCl_2 – zinc (II) chloride
7. Li_3N – lithium nitride
8. BaSO_3 – barium sulfite
9. Co_2S_3 – cobalt (III) sulfide
10. CsF – cesium fluoride
11. NaNO_3 – sodium nitrate
12. CuI_2 – copper (II) iodide
13. TiO_2 – titanium (IV) oxide
14. KMnO_4 – potassium permanganate
15. ZnCN_2 – zinc (II) cyanide
16. $\text{Al}_2(\text{CO}_3)_3$ – aluminum carbonate
17. $\text{Fe}_2(\text{CO}_3)_3$ – iron (III) carbonate
18. BaCrO_4 – barium chromate
19. $\text{Pb}_3(\text{PO}_4)_2$ – lead (II) phosphate
20. NaHCO_3 – sodium hydrogen carbonate or sodium bicarbonate

Write the formulas for the following compounds:

1. cesium bromide – CsBr
2. aluminum chlorite – $\text{Al}(\text{ClO}_2)_3$
3. ammonium phosphite – $(\text{NH}_4)_3\text{PO}_3$
4. copper (II) nitrate – $\text{Cu}(\text{NO}_3)_2$
5. sodium fluoride – NaF
6. lead (IV) sulfate – $\text{Pb}(\text{SO}_4)_2$
7. calcium carbonate – CaCO_3
8. barium oxide – BaO
9. iron (III) sulfide – Fe_2S_3
10. manganese (II) chromate – MnCrO_4

Formula Of Ionic Compound Lab Answer Key

AW Rasmussen



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Table of Contents Formula Of Ionic Compound Lab Answer Key

1. Understanding the eBook Formula Of Ionic Compound Lab Answer Key
 - The Rise of Digital Reading Formula Of Ionic Compound Lab Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Formula Of Ionic Compound Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Formula Of Ionic Compound Lab Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Formula Of Ionic Compound Lab Answer Key
 - Personalized Recommendations
 - Formula Of Ionic Compound Lab Answer Key User Reviews and Ratings
 - Formula Of Ionic Compound Lab Answer Key and Bestseller Lists
5. Accessing Formula Of Ionic Compound Lab Answer Key Free and Paid eBooks

- Formula Of Ionic Compound Lab Answer Key Public Domain eBooks
 - Formula Of Ionic Compound Lab Answer Key eBook Subscription Services
 - Formula Of Ionic Compound Lab Answer Key Budget-Friendly Options
6. Navigating Formula Of Ionic Compound Lab Answer Key eBook Formats
- ePub, PDF, MOBI, and More
 - Formula Of Ionic Compound Lab Answer Key Compatibility with Devices
 - Formula Of Ionic Compound Lab Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Formula Of Ionic Compound Lab Answer Key
 - Highlighting and Note-Taking Formula Of Ionic Compound Lab Answer Key
 - Interactive Elements Formula Of Ionic Compound Lab Answer Key
8. Staying Engaged with Formula Of Ionic Compound Lab Answer Key
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Formula Of Ionic Compound Lab Answer Key
9. Balancing eBooks and Physical Books Formula Of Ionic Compound Lab Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Formula Of Ionic Compound Lab Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Formula Of Ionic Compound Lab Answer Key
- Setting Reading Goals Formula Of Ionic Compound Lab Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Formula Of Ionic Compound Lab Answer Key
- Fact-Checking eBook Content of Formula Of Ionic Compound Lab Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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