

Chapter Three: Basic Principles of Chemistry

EXERCISES AND PROBLEMS

Problem 3.1

1. What was Mendeleev's approach to the periodicity of chemical properties?
Mendeleev rearranged the known elements. He placed each new element (those not yet discovered) in a column (group) vertically adjacent to elements with similar properties. He also placed elements with similar chemical properties in a row (period). He placed the noble gases in a column to the far right of the periodic table. By leaving an empty space adjacent to each element, Mendeleev predicted the existence of elements that had not yet been discovered. He predicted the atomic weights of these elements and the formulas of many of their compounds. He predicted the chemical and physical properties of these elements. Mendeleev's periodic table is the basis of the periodic table of elements that we use today.
2. What is the periodic law?
The periodic law states that the properties of the elements are a periodic function of their atomic number. The periodic law is the basis of the periodic table of elements.

Problem 3.2

1. What is the periodic law?
The periodic law states that the properties of the elements are a periodic function of their atomic number. The periodic law is the basis of the periodic table of elements.
2. How was Mendeleev's periodic table based on the periodic law?
Mendeleev's periodic table was based on the periodic law. He placed elements with similar properties in a column (group) vertically adjacent to elements with similar properties. He placed the noble gases in a column to the far right of the periodic table. By leaving an empty space adjacent to each element, Mendeleev predicted the existence of elements that had not yet been discovered. He predicted the atomic weights of these elements and the formulas of many of their compounds. He predicted the chemical and physical properties of these elements. Mendeleev's periodic table is the basis of the periodic table of elements that we use today.
3. What is the periodicity of chemical properties?
The periodicity of chemical properties is the periodicity of the chemical properties of the elements. The periodicity of chemical properties is the basis of the periodic table of elements.

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