

Chemistry Study Guide

Periodic Table of the Elements

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K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	Cs	Ba											Rn																																																			
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Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

Matter:

Anything that has **mass** and **occupies space** 🌟🌍

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the **smallest, indivisible particle** 🔬👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	$\frac{1}{1836}$	-1	Orbit/Shell

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Table of Contents Genral Chemistry Study Guide

1. Understanding the eBook Genral Chemistry Study Guide
 - The Rise of Digital Reading Genral Chemistry Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Genral Chemistry Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Genral Chemistry Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Genral Chemistry Study Guide
 - Personalized Recommendations
 - Genral Chemistry Study Guide User Reviews and Ratings
 - Genral Chemistry Study Guide and Bestseller Lists
5. Accessing Genral Chemistry Study Guide Free and Paid eBooks

- Genral Chemistry Study Guide Public Domain eBooks
 - Genral Chemistry Study Guide eBook Subscription Services
 - Genral Chemistry Study Guide Budget-Friendly Options
6. Navigating Genral Chemistry Study Guide eBook Formats
- ePub, PDF, MOBI, and More
 - Genral Chemistry Study Guide Compatibility with Devices
 - Genral Chemistry Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Genral Chemistry Study Guide
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 - Interactive Elements Genral Chemistry Study Guide
8. Staying Engaged with Genral Chemistry Study Guide
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Genral Chemistry Study Guide
9. Balancing eBooks and Physical Books Genral Chemistry Study Guide
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Genral Chemistry Study Guide
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Genral Chemistry Study Guide
- Setting Reading Goals Genral Chemistry Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Genral Chemistry Study Guide
- Fact-Checking eBook Content of Genral Chemistry Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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