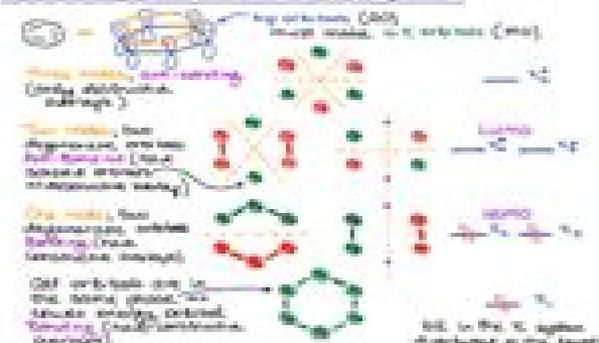


### Molecular Orbital in Benzene, Cyclohexadiene



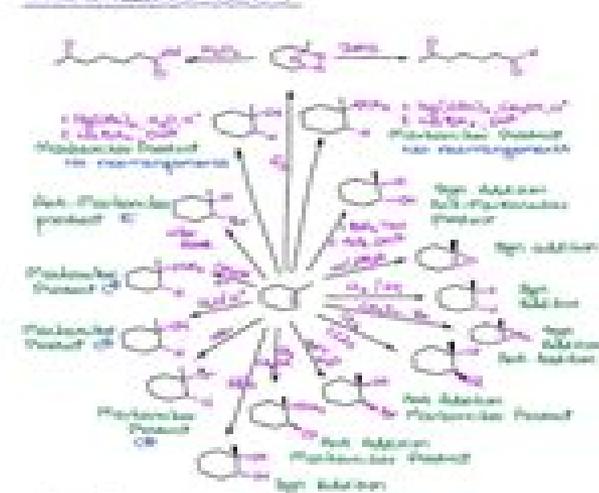
**Front-Back Diagrams** - show why pi bonds are formed between orbitals that are in phase.



### 3. Reduction of Alkyne with H<sub>2</sub> in Liquid Ammonia



### Alkyne Reaction network

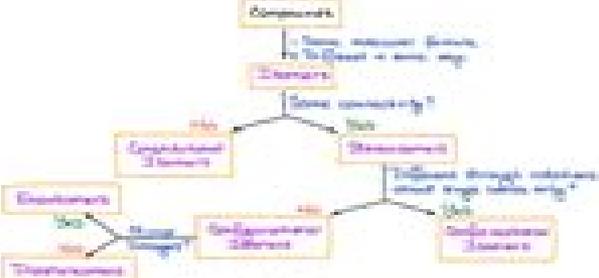


### Alkyne Chemical Group Comments



### Geometry of Stereoisomeric Relationships

- Constitutional Isomers** - completely different molecules
- Conformational Isomers** - same molecule that only differ in 3D shape due to free rotation around single bonds
- Enantiomers** - non-superimposable mirror images that share physical & chemical properties in achiral environment
- Diastereomers** - non-superimposable, non-mirror images that have different physical & chemical properties
- Meso Compounds** - non-chiral due to the plane of symmetry, however, have chiral centers



### Stereochemical Relationships of Structures

Enantiomers share identical physical & chemical properties in achiral environment, they can be isolated by chiral resolving agents.

Diastereomers have different physical & chemical properties.

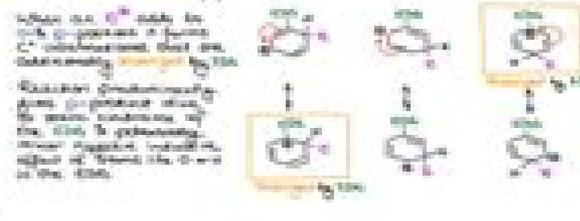
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### Electrophilic Aromatic Substitution Reactions (E<sub>1</sub>C<sub>2</sub>)

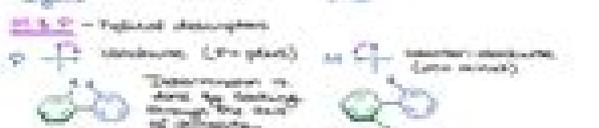


**Example:** Nitration of benzene.

The attack position depends on the steric effect of other groups present on the aromatic compound.



### Alkyne & Alkene - Markovnikov's Addition



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**JR Anderson**



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## **Table of Contents Guide Quick Notes Reactions Iitjee Organic Chemistry**

1. Understanding the eBook Guide Quick Notes Reactions Iitjee Organic Chemistry
  - The Rise of Digital Reading Guide Quick Notes Reactions Iitjee Organic Chemistry
  - Advantages of eBooks Over Traditional Books
2. Identifying Guide Quick Notes Reactions Iitjee Organic Chemistry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Guide Quick Notes Reactions Iitjee Organic Chemistry
  - User-Friendly Interface
4. Exploring eBook Recommendations from Guide Quick Notes Reactions Iitjee Organic Chemistry
  - Personalized Recommendations
  - Guide Quick Notes Reactions Iitjee Organic Chemistry User Reviews and Ratings
  - Guide Quick Notes Reactions Iitjee Organic Chemistry and Bestseller Lists

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  - Interactive Elements Guide Quick Notes Reactions Iitjee Organic Chemistry
8. Staying Engaged with Guide Quick Notes Reactions Iitjee Organic Chemistry
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Guide Quick Notes Reactions Iitjee Organic Chemistry
9. Balancing eBooks and Physical Books Guide Quick Notes Reactions Iitjee Organic Chemistry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Guide Quick Notes Reactions Iitjee Organic Chemistry
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Guide Quick Notes Reactions Iitjee Organic Chemistry
  - Setting Reading Goals Guide Quick Notes Reactions Iitjee Organic Chemistry
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Guide Quick Notes Reactions Iitjee Organic Chemistry
  - Fact-Checking eBook Content of Guide Quick Notes Reactions Iitjee Organic Chemistry
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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