

10 • Gas Laws

UNIT REVIEW

Directions: Use the following questions to guide your studying. Also, review your notes packet and look at questions you missed on quizzes and homework. Doing this study guide does not guarantee that you will ace the test. You need to ask questions if you have them!

Useful Information

STP = 0°C = 273 K and
 1 atm = 760 torr = 760 mmHg
 = 101.3 kPa = 14.7 psi
 = 14.7 $\frac{\text{lb}}{\text{in}^2}$

Ideal Gas Constant, R
 = 62.4 $\frac{\text{L}\cdot\text{mmHg}}{\text{mol}\cdot\text{K}}$
 = 0.0821 $\frac{\text{L}\cdot\text{atm}}{\text{mol}\cdot\text{K}}$
 = 8.31 $\frac{\text{L}\cdot\text{kPa}}{\text{mol}\cdot\text{K}}$

1. Perform the following conversions.

a. $26.0^\circ\text{C} = \underline{299 \text{ K}}$
 $26.0^\circ\text{C} + 273 = 299$

b. $135^\circ\text{C} = \underline{408 \text{ K}}$
 $135^\circ\text{C} + 273 = 408$

c. $-127^\circ\text{C} = \underline{146 \text{ K}}$
 $-127^\circ\text{C} + 273 = 146 \text{ K}$

d. $100 \text{ K} = \underline{-173}^\circ\text{C}$
 $100 \text{ K} = x + 273$, $x = -173$

e. $400 \text{ K} = \underline{127}^\circ\text{C}$ $400 \text{ K} = x + 273$
 $x = 127$

f. $4 \text{ K} = \underline{-269}^\circ\text{C}$
 $4 = x + 273$
 $x = -269$

2. What is the temperature of a sample of gas that has double the kinetic energy (energy of motion) of a sample at 80 °C.

$80^\circ\text{C} = 353 \text{ K}$

Kelvin temperature
 measures kinetic energy
 If you double the
 kinetic energy, then
 you double the Kelvins.
 $2(353 \text{ K}) = 706 \text{ K}$

Gas Laws Unit 9 Chemistry Review Key

David Baud



Gas Laws Unit 9 Chemistry Review Key:

Decoding **Gas Laws Unit 9 Chemistry Review Key**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Gas Laws Unit 9 Chemistry Review Key**," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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