

## Chemistry: General Equilibria and Kinetics Questions

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### Equilibrium and Kinetics (View all 200 questions) (View all 200 questions)

1. A sample of gas at 1.00°C and 1.00 atmosphere pressure is heated. Which volume will show gas expansion at 100°C and 1.00 atm?
2. A sample of hydrogen gas at 10.0 L, at a temperature of 100°C and a pressure of 1.00 atm. Calculate the number of H<sub>2</sub> molecules present. Write your answer.
3. An amount of water at 100°C and at a pressure of 1.00 atm is heated. Which effect does gas expansion have on the volume of the gas? Which effect does the pressure of the gas have on the volume? Which effect does temperature have on the volume?
4. A 1.00 L gas sample at 100°C and a pressure of 1.00 atm is heated to a volume of 2.00 L. What final pressure of the gas is at 100°C? Write your final temperature of the gas in Kelvin.
5. What is the volume of 1.00 mole of H<sub>2</sub> at 100°C?
6. A 1.00 mole of gas at a volume of 1.00 L and a temperature of 100°C. What is the final volume of the gas at a pressure of 1.00 atm and 100°C? Write your final volume of the gas.
7. A 1.00 mole of nitrogen gas at a volume of 10.0 L, at 100°C. What volume will 2.00 mole of nitrogen gas occupy at the same temperature and pressure?
8. A 1.00 L container filled with 1.00 g of gas. What is the temperature of the gas at 1.00 atm?
9. A gas sample is heated by 1.00 mole of gas. The volume of the gas is increased to 1.00 L. Is the volume of the gas 1.00 L, or is it more than 1.00 L? Is the volume of the gas 1.00 L, or is it less than 1.00 L? Is the volume of the gas 1.00 L, or is it more than 1.00 L? Write your final volume of the gas.
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15. A gas sample is heated by 1.00 mole of gas. The volume of the gas is increased to 1.00 L. What is the final pressure of the gas at 1.00 atm? Write your final pressure of the gas.

# Gases Study Guide Answers

**F Rizvi**



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