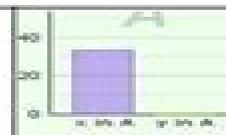


Activity A:
Temperature and diffusion

Get the Gizmo ready:

- Click **Reset** (↺).
- Set the **Wall** to 100%.



Introduction: In this Gizmo, temperature is measured on the **Kelvin scale**. On this scale, 0 K represents **absolute zero**, the coldest possible temperature. Water freezes at 273.15 K (0 °C), and water boils at 373.15 K (100 °C).

Question: How does temperature affect the rate of diffusion?

1. **Observe:** Click **Reset**. Set the temperature to 100 K, and click **Play**. Observe the number of particles that pass through the wall. Then set the temperature to 200 K, and **Play** again.
- How does the temperature of the gas relate to the number of particles that pass through the wall?

As the temperature of the gas increases, the number of particles that pass through the wall increases.

The number of particles that pass through the wall is a measure of the average **kinetic energy** of the particles. Kinetic energy (KE) depends on the velocity and mass of the particles. $KE = \frac{1}{2}mv^2$.

2. **Observe:** Click **Reset**. How do you think temperature will affect the rate of diffusion?

As the temperature increases, the rate of diffusion will also increase.

3. **Observe:** Click **Reset**. Set the wall to 100% and click **Play**. Set the temperature to 100 K, and **Play** again. Then set the temperature to 200 K, and **Play** again.

Click **Play** until you see at least 100 particles on each side. Record the time it takes for 100 particles to pass through the wall on each side.

Repeat this procedure at 100 K, and then at the wall with the temperature set to 200 K.

Temp = 100 K (approximate)

Temp	Time to reach equilibrium
100 K	1000
200 K	500

Explore Learning Gizmo Answer Key Diffusion

JA Banks



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Table of Contents Explore Learning Gizmo Answer Key Diffusion

1. Understanding the eBook Explore Learning Gizmo Answer Key Diffusion
 - The Rise of Digital Reading Explore Learning Gizmo Answer Key Diffusion
 - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Gizmo Answer Key Diffusion
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Gizmo Answer Key Diffusion
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Gizmo Answer Key Diffusion
 - Personalized Recommendations
 - Explore Learning Gizmo Answer Key Diffusion User Reviews and Ratings
 - Explore Learning Gizmo Answer Key Diffusion and Bestseller Lists
5. Accessing Explore Learning Gizmo Answer Key Diffusion Free and Paid eBooks
 - Explore Learning Gizmo Answer Key Diffusion Public Domain eBooks
 - Explore Learning Gizmo Answer Key Diffusion eBook Subscription Services
 - Explore Learning Gizmo Answer Key Diffusion Budget-Friendly Options

6. Navigating Explore Learning Gizmo Answer Key Diffusion eBook Formats
 - ePub, PDF, MOBI, and More
 - Explore Learning Gizmo Answer Key Diffusion Compatibility with Devices
 - Explore Learning Gizmo Answer Key Diffusion Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Gizmo Answer Key Diffusion
 - Highlighting and Note-Taking Explore Learning Gizmo Answer Key Diffusion
 - Interactive Elements Explore Learning Gizmo Answer Key Diffusion
8. Staying Engaged with Explore Learning Gizmo Answer Key Diffusion
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Gizmo Answer Key Diffusion
9. Balancing eBooks and Physical Books Explore Learning Gizmo Answer Key Diffusion
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Gizmo Answer Key Diffusion
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Explore Learning Gizmo Answer Key Diffusion
 - Setting Reading Goals Explore Learning Gizmo Answer Key Diffusion
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Explore Learning Gizmo Answer Key Diffusion
 - Fact-Checking eBook Content of Explore Learning Gizmo Answer Key Diffusion
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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