

Student Exploration: Ionic Bonds

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Nate can give Clara a marker

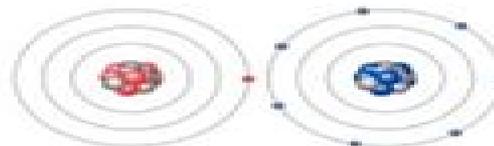
2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggie can give Fred and Florence one marker each

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The Ionic Bonds Gizmo™ allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** (▶) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain electrons. The outermost electrons are called **valence electrons**. (Inner electrons are not shown.)

How many valence electrons does each atom have?

Sodium: 1

Chlorine: 7

2. Click **Pause** (⏸). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

sodium

Which is a nonmetal?

chlorine

B. Try moving an electron from the metal to the nonmetal. What happens?

The chlorine atom took the electron and added it to its outermost shell

Explore Learning Ionic Bonds Answers

S Ashworth



Explore Learning Ionic Bonds Answers:

This book delves into Explore Learning Ionic Bonds Answers. Explore Learning Ionic Bonds Answers is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Explore Learning Ionic Bonds Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Explore Learning Ionic Bonds Answers
 - Chapter 2: Essential Elements of Explore Learning Ionic Bonds Answers
 - Chapter 3: Explore Learning Ionic Bonds Answers in Everyday Life
 - Chapter 4: Explore Learning Ionic Bonds Answers in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Explore Learning Ionic Bonds Answers. This chapter will explore what Explore Learning Ionic Bonds Answers is, why Explore Learning Ionic Bonds Answers is vital, and how to effectively learn about Explore Learning Ionic Bonds Answers.
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 4. In chapter 3, this book will examine the practical applications of Explore Learning Ionic Bonds Answers in daily life. This chapter will showcase real-world examples of how Explore Learning Ionic Bonds Answers can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Explore Learning Ionic Bonds Answers in specific contexts. The fourth chapter will explore how Explore Learning Ionic Bonds Answers is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Explore Learning Ionic Bonds Answers. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Explore Learning Ionic Bonds Answers.

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