

Student Exploration: Phase Changes

Approximate time: 45 minutes

Pages: 13 (with Student Answer Key)

Version: 1.0 (2014)

Prerequisite Knowledge: Students should have completed the following activities:

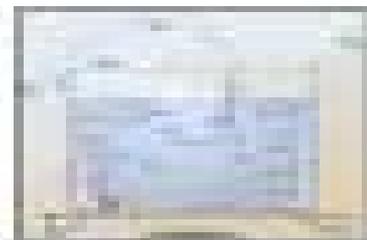
- 1. [Energy Transfer: Conduction](#) (for mass, volume, and temperature) and [Energy Transfer: Convection](#) (for mass, volume, and temperature)

Introduction: In this activity, you will explore the phase changes of water.

Learning Objectives:

By the end of this activity, you should be able to:

- 1. Explain the phase changes of water and the energy transfer involved.
- 2. Explain the relationship between the phase change and the temperature of the water.
- 3. Explain the relationship between the phase change and the volume of the water.



1. What is the relationship between the phase change and the energy transfer?

Answer: Energy transfer is required for all phase changes.

2. What is the relationship between the phase change and the volume of the water?

Answer: The volume of the water changes during phase changes.

3. What is the relationship between the phase change and the temperature of the water?

Answer: The temperature of the water changes during phase changes.

4. What is the relationship between the phase change and the density of the water?

Answer: The density of the water changes during phase changes.

<p>Key Concepts:</p> <ul style="list-style-type: none"> • Phase changes occur when energy is added or removed from a substance. • The temperature of a substance remains constant during a phase change. • The volume of a substance changes during a phase change. 	
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Procedure: You will use the simulation to explore the phase changes of water.

1. **Introduction:** Use the simulation to explore the phase changes of water. When the simulation starts, the temperature of the water is 100°C.

Question: What happens to the temperature of the water as it changes phase? Record your observations in the table below.

- | | | |
|---|-----------------------------------|-------------------------------------|
| 1. As the temperature of the water decreases from 100°C to 0°C, what happens to the temperature of the water? | It stays constant at 100°C. | This is the boiling point . |
| 2. As the temperature of the water decreases from 0°C to -10°C, what happens to the temperature of the water? | It stays constant at 0°C. | This is the freezing point . |
| 3. As the temperature of the water decreases from -10°C to -20°C, what happens to the temperature of the water? | It decreases from -10°C to -20°C. | This is the freezing point . |

2. **Boiling:** Use the simulation to explore the boiling point of water. When the simulation starts, the temperature of the water is 100°C.

Question: What happens to the temperature of the water as it boils? Record your observations in the table below.

3. **Freezing:** Use the simulation to explore the freezing point of water. When the simulation starts, the temperature of the water is 0°C.

Question: What happens to the temperature of the water as it freezes? Record your observations in the table below.



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