

# Lab #8: Exploring the Properties of Gases

## **Purpose**

In this experiment you will explore three gas laws Boyle's Law, Charles's Law, and Gay-Lussac's Law also known as Amontons's Law. The first activity relates pressure and volume. The second activity relates temperature and volume. The third activity relates temperature and pressure. Data will be gathered for each gas law, plotted as graphs and the necessary calculations will be completed to evaluate the gas law in each activity.

## **Introduction**

The purpose of this investigation is to conduct three activities to study the empirical gas laws, Boyle's Law, Amontons's Law, and Charles's Law. Three properties of gases will be investigated: pressure, volume, and temperature. You will be using an online simulation to change the one of these variables and examine the effect on the others. By gathering and analyzing your data and observations, you will be able to describe the gas laws, both qualitatively and mathematically.

### ***Boyle's Law***

According to Boyle's Law, the pressure (P) of a gas varies inversely with the volume (V) of the gas when the temperature (T) and moles of gas (n) are kept constant.

$$P \propto \frac{1}{V} \quad (T \text{ and } n \text{ are constant})$$

According to Boyle's Law, volume decreases when the pressure increases, and volume increases when the pressure decreases.

$$P_2 \times V_2 = P_1 \times V_1$$

In this lab you will start with a known volume of gas and a pressure sensor will measure the pressure of the gas. You will decrease the volume of gas in increments, each time measuring the new pressure of the gas.

### ***Amontons's Law***

According to Amontons's Law, the pressure of a gas varies proportionally with the Kelvin temperature as long as the volume and quantity of gas are kept constant.

# Exploring The Properties Of Gases Lab Answers

**Rosina Ehmman**



## Exploring The Properties Of Gases Lab Answers:

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Research and Development Administration. Technical Information Center,1976  
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In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Exploring The Properties Of Gases Lab Answers**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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