

THE ULTIMATE GENERAL CHEMISTRY 1 STUDY GUIDE

❖ Matter and Measurement

- ✔ Matter: anything that has mass and occupies space
- ✔ Properties of matter: physical properties and chemical properties
- ✔ States of matter: solid, liquid, gas
- ✔ Measurements: SI Units, significant figures, dimensional analysis

❖ Atoms, Molecules, and Ions

- ✔ Atomic structure: protons, neutrons, and electrons
- ✔ Periodic table and trends: electronegativity, ionization energy, electron affinity
- ✔ Chemical bonding: ionic bonding, covalent bonding, metallic bonding
- ✔ Chemical formulas and equations: empirical and molecular formulas, balancing chemical equations

❖ Stoichiometry

- ✔ Limiting reactants and percent yield
- ✔ Mole concept and Avogadro's number
- ✔ Molarity and molality
- ✔ Gas laws: Boyle's law, Charles's law, Gay-Lussac's law, Avogadro's law, Ideal gas law

❖ Chemical Reactions and Equations

- ✔ Types of chemical reactions: synthesis, decomposition, single replacement, double replacement, combustion
- ✔ Redox reactions and balancing redox equations
- ✔ Reaction rates: factors affecting reaction rates, rate laws, and reaction mechanisms

❖ Gases

- ✔ Gas laws: Boyle's law, Charles's law, Gay-Lussac's law, Avogadro's law, Ideal gas law
- ✔ Kinetic Molecular Theory of gases: pressure, temperature, volume, and the number of moles of gas particles
- ✔ Natural gases: van der Waals equation

General Chemistry Thinking Guide

L Reisser



General Chemistry Thinking Guide:

Catalog of Copyright Entries, Third Series Library of Congress. Copyright Office, 1977 Introductory Chemistry Neil Elbridge Gordon, 1927 *Resources in Education* , 1984 **Chemists' Guide to Effective Teaching** Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry Handbook of the Undergraduate Curriculum Jerry G. Gaff, James L. Ratcliff, 1997 This volume offers a compAndium of the best ideas analyses and practices relating to the undergraduate curriculum as described by leading figures in the field It contains both conceptual and practical information on effective practices research management and assessment In thirty four original chapters top practitioners and scholars detail a range of philosophies frameworks program designs instructional strategies and assessment methods being used to strengthen and transform the curriculum They examine both the current state of knowledge and teaching in the disciplines and the forces that will reshape the curriculum in the coming years The Handbook of Undergraduate Curriculum will prove valuable both to practitioners as an operating manual or desk reference and to faculty as a primary text for graduate courses on the curriculum In addition the book will be a useful tool for those serving on a general education curriculum committee or conducting a departmental review of a major program as well as having numerous other practical applications for anyone with responsibility for or interest in the curriculum Bibliographic Guide to Education , 1989 **General Chemistry** Darrell D. Ebbing, Steven D. Gammon, 2008-02 The perfect complement to the main text the Study Guide further emphasizes the step by step problem solving approach with an emphasis on the thinking processes required Each chapter contains a list of key terms and their definitions learning objectives additional worked out exercises tied to the in chapter examples a diagnostic test summaries of major concepts additional practice problems and a post test Eight additional cumulative unit exams are provided Helpful tips on how to use the textbook study effectively and succeed in general chemistry make this a valuable resource for students

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