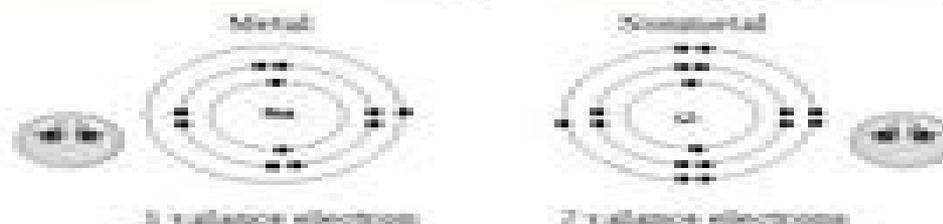


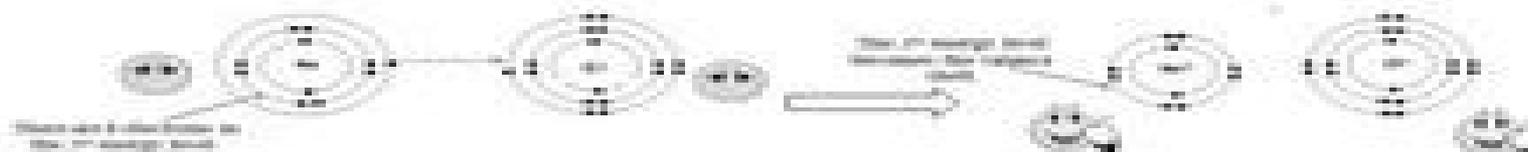
- The picture below shows the Bohr model of an "unhappy" metal atom and the Bohr model of an "unhappy" nonmetal atom.



- How do you think these two atoms could work together to achieve happiness (8 valence electrons)?

- Look at the illustration below.

This illustration shows how a sodium and a chlorine atom can work together to form stable electron shells with eight valence electrons.



We can summarize this process by using Lewis dot structures as shown below:



- The sodium atom donated its one valence electron to the chlorine atom. How did this action cause the sodium atom to end up with eight valence electrons?

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