

# face2face Placement Test

## Question Sheet

- Choose the best answer for each question.
- Choose one answer only.
- Write your answers on the Answer Sheet.
- Spend a maximum time of 30 minutes on the test.
- Please stop if the questions become too difficult.

- 1 Maria \_\_\_\_\_ Brazilian.  
a) are    b) am    c) is    d) be
- 2 Paul's studying medicine because he wants to be a \_\_\_\_\_.  
a) lawyer    b) teacher    c) doctor    d) builder
- 3 Jim \_\_\_\_\_ got a car.  
a) doesn't    b) hasn't    c) isn't    d) haven't
- 4 John is \_\_\_\_\_ brother.  
a) Peter's    b) Peter    c) of Peter    d) Peter his
- 5 What's your \_\_\_\_\_ sport?  
a) important    b) favourite    c) beautiful    d) nice
- 6 Excuse me. \_\_\_\_\_ time please?  
a) You have the    b) What is    c) Have you got the    d) What
- 7 What time \_\_\_\_\_ get up in the morning?  
a) you    b) do you    c) are you    d) you do
- 8 I usually \_\_\_\_\_ at the supermarket on Saturday mornings.  
a) go for a drink    b) eat out    c) do sport    d) go shopping
- 9 What \_\_\_\_\_ Mary for her birthday?  
a) about buying    b) shall we buy    c) to buy    d) let's
- 10 My brother \_\_\_\_\_ in London.  
a) live    b) lives    c) is lived    d) living
- 11 \_\_\_\_\_ there a restaurant near here?  
a) Are    b) Have    c) Do    d) Is
- 12 Do you prefer to have a \_\_\_\_\_ or a bath in your bathroom?  
a) shower    b) cooker    c) washing machine    d) fridge

# Face2face Placement Test

**Kathleen Armour**



## Face2face Placement Test:

face2face Advanced Teacher's Book with DVD Theresa Clementson, Gillie Cunningham, Jan Bell, 2013-09-26 face2face  
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In a world inundated with displays and the cacophony of fast communication, the profound power and emotional resonance of verbal art usually diminish in to obscurity, eclipsed by the regular barrage of noise and distractions. Yet, located within the musical pages of **Face2face Placement Test**, a fascinating function of fictional beauty that impulses with raw feelings, lies an unique journey waiting to be embarked upon. Penned by way of a virtuoso wordsmith, that enchanting opus manuals visitors on a mental odyssey, lightly exposing the latent potential and profound influence stuck within the complex web of language. Within the heart-wrenching expanse with this evocative analysis, we shall embark upon an introspective exploration of the book is central subjects, dissect its charming publishing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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jan 3 2023 a net ionic equation is a chemical equation that shows only the species participating in a chemical reaction and omits the spectator ions spectator ions are ions that

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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations

requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction

therefore the net ionic equation is  $2\text{Cl}^- + 2\text{Ag}^+$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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may 11 2017 in the net ionic equation any ions that do not participate in the reaction called spectator ions are excluded as a result the net ionic equation shows only the species that

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1  $\text{K}_2\text{PO}_4$  aq +  $\text{Al}(\text{NO}_3)_3$  aq →  $3\text{KNO}_3$  aq +  $3\text{AlPO}_4$  s  
 3  $\text{Mg}(\text{NO}_3)_2$  aq +  $3\text{Mg}(\text{OH})_2$  s  
 aq +  $3\text{Mg}(\text{OH})_2$  s

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oct 2 2019 the net ionic equation is a chemical equation for a reaction that lists only those species participating in the reaction the net ionic equation is commonly used in acid base

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jun 23 2022 what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic

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aug 10 2023 write the full ionic and net ionic equations for this reaction solution the elemental metals magnesium on the reactant side copper on the product side are neutral

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds cl aq

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh<sub>4</sub> aq cr<sub>2</sub>o<sub>7</sub><sup>2-</sup> aq ag<sub>2</sub>cr<sub>2</sub>o<sub>7</sub> s 2nh<sub>4</sub> aq 2f aq because two nh<sub>4</sub> aq

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