

Name: _____

Date: _____

Period: _____

Atomic Structure, Valence Electrons, and the Periodic Table

Background Info:

Atoms are the smallest unit consist of three main subatomic particles: positively charged particles (protons) with neutral neutrons. Negatively charged electrons orbit the nucleus in shells. An atom's electrons determine its chemical reactivity.

Periodic Table of Elements:

Label the Periods, Families, and Valence Electrons.

1	2				
3	4				
5	6	7			
8	9	10	11	12	13
14	15	16	17	18	19
20	21	22	23	24	25
26	27	28	29	30	31
32	33	34	35	36	37

38	39	40	41	42	43
44	45	46	47	48	49

Valence Electrons:

Write the Number of Valence Electrons for each element.

- S = _____ F = _____
- H = _____ Cl = _____
- Al = _____ I = _____

Reading Elements on the Periodic Table:

Fill in the Information Below

4

Bc

Beryllium

9.0122

- ← _____, Number of Protons (and Electrons)
- ← _____, Element Abbreviation
- ← _____, Formal Element Name
- ← _____, Equal to Protons + Neutrons

- How to Find Protons? → Look at the Atomic Number: Atomic # _____ Protons # _____
- How to Find Electrons? → Look at the Atomic Number: Atomic # _____ Electrons # _____
- How to Find Neutrons? → Take Mass # - Atomic #: Mass # _____ Neutrons # _____

Reading Elements on the Periodic Table:

Fill in the Information Below using the Rounded Mass Number

18

Ar

Argon

39.948

- Name: _____
- Symbol: _____
- Protons: _____
- Neutrons: _____
- Electrons: _____
- Val. e⁻: _____

15

P

Phosphorus

30.974

- Name: _____
- Symbol: _____
- Protons: _____
- Neutrons: _____
- Electrons: _____
- Val. e⁻: _____

31

Ga

Gallium

69.723

- Name: _____
- Symbol: _____
- Protons: _____
- Neutrons: _____
- Electrons: _____
- Val. e⁻: _____

53

I

Iodine

126.90

- Name: _____
- Symbol: _____
- Protons: _____
- Neutrons: _____
- Electrons: _____
- Val. e⁻: _____

52

Te

Tellurium

127.60

- Name: _____
- Symbol: _____
- Protons: _____
- Neutrons: _____
- Electrons: _____
- Val. e⁻: _____

55

Cs

Cesium

132.91

- Name: _____
- Symbol: _____
- Protons: _____
- Neutrons: _____
- Electrons: _____
- Val. e⁻: _____

1

H

Hydrogen

1.008

- Name: _____
- Symbol: _____
- Protons: _____
- Neutrons: _____
- Electrons: _____
- Val. e⁻: _____

12

Mg

Magnesium

24.305

- Name: _____
- Symbol: _____
- Protons: _____
- Neutrons: _____
- Electrons: _____
- Val. e⁻: _____

Huntinh The Elements Packet Answers

Daniela Niemeyer



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Table of Contents Huntinh The Elements Packet Answers

1. Understanding the eBook Huntinh The Elements Packet Answers
 - The Rise of Digital Reading Huntinh The Elements Packet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Huntinh The Elements Packet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Huntinh The Elements Packet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Huntinh The Elements Packet Answers
 - Personalized Recommendations
 - Huntinh The Elements Packet Answers User Reviews and Ratings
 - Huntinh The Elements Packet Answers and Bestseller Lists
5. Accessing Huntinh The Elements Packet Answers Free and Paid eBooks
 - Huntinh The Elements Packet Answers Public Domain eBooks
 - Huntinh The Elements Packet Answers eBook Subscription Services
 - Huntinh The Elements Packet Answers Budget-Friendly Options
6. Navigating Huntinh The Elements Packet Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Huntinh The Elements Packet Answers Compatibility with Devices
 - Huntinh The Elements Packet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Huntinh The Elements Packet Answers
 - Highlighting and Note-Taking Huntinh The Elements Packet Answers
 - Interactive Elements Huntinh The Elements Packet Answers
 8. Staying Engaged with Huntinh The Elements Packet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Huntinh The Elements Packet Answers
 9. Balancing eBooks and Physical Books Huntinh The Elements Packet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Huntinh The Elements Packet Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Huntinh The Elements Packet Answers
 - Setting Reading Goals Huntinh The Elements Packet Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Huntinh The Elements Packet Answers
 - Fact-Checking eBook Content of Huntinh The Elements Packet Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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