



PARTICLE THEORY AND STATES OF MATTER

1 What is particle theory?

2 All of matter is made of particles. Explain in solids and liquids?

3 Describe how particles move in a liquid. Explain why they are not fixed in place?

4 The particles in a gas have high kinetic energy. Explain how this is different from solids and liquids?

5 How are particles arranged in a crystal?

11 Explain how a substance can change from a solid to a liquid and then to a gas. What is the change of state called? Give one example of a substance that can do this!

Particle theory states that all matter is made up of tiny particles. These particles have three main properties: they are constantly moving, they are very small, and they have spaces between them. The arrangement and movement of particles depend on the state of matter. In solids, particles are packed closely together and vibrate in fixed positions. In liquids, they are more disordered and can move past each other. In gases, they are far apart and move rapidly in all directions.

Diffusion: This is the spread of particles from an area of high concentration to an area of low concentration. It occurs in all states of matter but is fastest in gases. It happens because particles are constantly moving and colliding.

Expansion of Solids: When a solid is heated, its particles vibrate more vigorously. This causes the solid to expand because the particles push each other further apart.

Change of State: A change of state occurs when the temperature of a substance reaches a certain point. For example, melting occurs when a solid is heated enough for its particles to overcome the forces holding them together.

Energy of Particles: Particles in a solid have low kinetic energy, particles in a liquid have medium energy, and particles in a gas have high kinetic energy. This energy is related to the temperature of the substance.

Pressure: Pressure is exerted by particles in a gas because they are constantly moving and colliding with the walls of their container.

Crystallization: This is the process where a liquid turns into a solid crystal. It happens when the temperature of the liquid drops below its freezing point.

Evaporation: This is the process where a liquid turns into a gas. It happens when particles at the surface of a liquid gain enough energy to escape into the air.



6 How do the particles in a gas move? Give one example of a gas.

10 Describe how particles in a solid are arranged. How do they move? Give one example of a solid.

9 What is the change of state called when a solid turns into a liquid?

8 Describe how particles in a liquid are arranged. How do they move? Give one example of a liquid.

7 How do the particles in a gas move? Give one example of a gas.

Particle Theory and States of Matter Guided Reading

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Brinkley, James M. McPherson**

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