

Solubility Values

- Amount of substance required to form a saturated solution with a specific amount of solvent at a specified temperature. Gases must also have pressure specified.

TABLE 13-4 Solubility of Solutes as a Function of Temperature (In g solute/100. g H₂O)

Substance	Temperature (°C)					
	0	20	40	60	80	100
AgNO ₃	122	216	311	440	585	733
Ba(OH) ₂	1.67	3.89	8.22	20.94	101.4	—
C ₁₂ H ₂₂ O ₁₁	179	204	238	287	362	487
Ca(OH) ₂	0.189	0.173	0.141	0.121	—	0.07
Ce ₂ (SO ₄) ₃	20.8	10.1	—	3.87	—	—
KCl	28.0	34.2	40.1	45.8	51.3	56.3
KI	128	144	162	176	192	206
KNO ₃	13.9	31.6	61.3	106	167	245
LiCl	69.2	83.5	89.8	98.4	112	128
Li ₂ CO ₃	1.54	1.33	1.17	1.01	0.85	0.72
NaCl	35.7	35.9	36.4	37.1	38.0	39.2
NaNO ₃	73	87.6	102	122	148	180
CO ₂ (gas at SP)	0.335	0.169	0.0973	0.058	—	—
O ₂ (gas at SP)	0.00694	0.00537	0.00308	0.00227	0.00138	0.00

Holt Chemistry Solubility And The Dissolving Process

H. Clark Metcalfe, Joseph F. Castka



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