

Solution Manual For Laboratory Experiments for Introduction to General, Organic and Biochemistry, 8th Edition

Experiment 1

This may be a student's first experience in the laboratory. Therefore, the instructor should demonstrate all the techniques used in this laboratory. Show how a Bunsen burner is lit, with a match or a gas striker, and how the flame is adjusted by control of the gas valve and air vents.

This is a relatively simple laboratory for students to work. Most of the common equipment used in the laboratory are introduced here. For many this might be the first time some of the glassware will be encountered. For the instructor, patience is in order since the lack of familiarity of the student with the laboratory ware often creates problems. Take the graduated cylinder, for example. Since it is tall, it is easily knocked over, and although laboratory glassware is reasonably durable, it will shatter and could cause severe cuts. Remind students not to pick up broken glass with the fingers but to use the dustpan and brush. Broken glass should be discarded in a waste container specifically for glass.

While there is little danger in this laboratory of eye damage, nevertheless, it is essential that the rules of the laboratory be followed: **safety glasses are to be worn at all times in the laboratory.**

The thermometers in this laboratory are made of glass and must be handled properly. A thermometer is not a stirring rod and must not be used as such. If a student wants to bring the fluid level in the thermometer down, remind him/her to use cold water from the tap. The laboratory thermometer is not a clinical thermometer and does not require that it be shaken down! Waving the thermometer usually results in it hitting a bench top and breaking. Some of these thermometers contain mercury; the breakage of a thermometer with resultant spillage of mercury must be cleaned up quickly. Mercury is toxic, especially as a vapor. The instructor should be notified immediately for proper clean up. No mercury should be left freely about anywhere. Mercury can be collected with commercial collectors or by a homemade suction apparatus. Connect a side-arm suction filter flask to a water aspirator. The flask is fitted with a one-hole rubber stopper with a small section of glass tubing inserted into the hole. Rubber tubing connects the glass tube to a Pasteur pipet. When the water is turned on, the spheres of mercury will be sucked into the pipet and then into the suction flask. The recovered mercury can be stored under water.

Balances should be handled with care; electronic top-loading balances are sensitive and lose calibration easily. Demonstrate proper use of the balance. Emphasize that no chemical should be weighed directly on the pan; use either weighing paper or a suitable container. Also

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AW Chickering



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