

## CHEM 105

### Module 1: Introduction to Chemistry and Matter

#### The Nature of Science and Chemistry

- Science: knowledge
  - Specifically, knowledge gained through experience or experiment – empirical knowledge.
  - Science is both an activity and the result of the activity.
- Chemistry: the study of matter and its interactions with other matter and with energy.

#### Chemistry is EVERYWHERE

- In the air we breathe.
- The food we eat.
- Colors we see.
- Fuels we use.
- Medicines we take.
- In all living organisms, like US.

#### What is Chemistry?

- The study of matter and the changes it can undergo.
  - o Matter is anything that has mass and occupies space.
- Chemistry plays an important central role in all of the natural sciences:
  - o Geology – Geo-chemistry
  - o Astronomy – Cosmo-chemistry
  - o Physics – Chemical physics
  - o Biology – biochemistry

#### The Scientific Method

- The scientific method is a list of techniques for investigating occurrences, acquiring new knowledge, or adjusting and integrating previous knowledge. This list includes:
  - o Observation- observe a phenomenon.
  - o Hypothesis- pose a question explaining the phenomenon.
  - o Experiments- run experiments to prove or disprove your hypothesis.
  - o Publication- allow outside reviews of your experiments.
  - o Confirmation- confirm your experimental results.
  - o Application - determine what scientific use your results have.

#### Types of Scientific Research

- Basic science- research without the goal of a practical application.
- Applied science- research with well-defined, short-term goals for a specific problem.
- Technology- application of science for industrial production and societal goals.

#### Types of Matter

- Macroscopic – large enough to be seen with the naked eye.
- Microscopic- seen only with the aid of a microscope.
- Submicroscopic- can only be observed indirectly.

#### Properties of Matter

- Mass is a measure of the amount or quantity of matter in an object.

# Introduction To Chemistry 105 Study Guide

**Barron's Educational Series, Terry L.  
Duran**



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**Resources in Education** , 1986     **Inorganic Chemistry** James E. House, 2012-12-31 Inorganic Chemistry Second Edition provides essential information for students of inorganic chemistry or for chemists pursuing self study The presentation of topics is made with an effort to be clear and concise so that the book is portable and user friendly The text emphasizes fundamental principles including molecular structure acid base chemistry coordination chemistry ligand field theory and solid state chemistry It is organized into five major themes structure condensed phases solution chemistry main group and coordination compounds with several chapters in each There is a logical progression from atomic structure to molecular structure to properties of substances based on molecular structures to behavior of solids etc The textbook contains a balance of topics in theoretical and descriptive chemistry For example the hard soft interaction principle is used to explain hydrogen bond strengths strengths of acids and bases stability of coordination compounds etc Discussion of elements begins with survey chapters focused on the main groups while later chapters cover the elements in greater detail Each chapter opens with narrative introductions and includes figures tables and end of chapter problem sets This new edition features new and improved illustrations including symmetry and 3D molecular orbital representations expanded coverage of spectroscopy instrumental techniques organometallic and bio inorganic chemistry and more in text worked out examples to encourage active learning and to prepare students for their exams This text is ideal for advanced undergraduate and graduate level students enrolled in the Inorganic Chemistry course This core course serves Chemistry and other science majors The book may also be suitable for biochemistry medicinal chemistry and other professionals who wish to learn more about this subject area Concise coverage maximizes student understanding and minimizes the inclusion of details students are unlikely to use Discussion of elements begins with survey chapters focused on the main groups while later chapters cover the elements in greater detail Each chapter opens with narrative introductions and includes figures tables and end of chapter problem sets

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