

Name: KEY

Matter Unit Test

Introduction To Matter Section 1 - Multiple Choice Questions are 4 points each

P 1. Matter is defined as any object that _____

- a. Has mass
- b. Has volume
- c. Is alive - *metals/salt, sugar not alive but matter*
- d. Has mass and volume (takes up space)

if it has mass, it must have volume

C 2. Which of the following statements about the periodic table is TRUE?

- a. All elements on the periodic table belong to the same family.
- b. Elements on the periodic table are organized by where they are found in nature.
- c. Elements on the periodic table are unique from each other, but do share common characteristics with other elements.
- d. Elements on the periodic table can be broken down into more simple substances.

weight

they cannot

B 3. A piece of iron and a piece of gold have the same mass. Which statement best describes the metals?

- a. The amount of matter in the iron is different than the amount of matter in the gold. *-not enough info*
- b. The number of atoms in the iron is different than the number of atoms in gold.
- c. The melting point of the iron is the same as the melting point of the gold. *different atoms*
- d. The volume of the iron will be the same as the volume of gold. *-not enough info*

B 4. If we were to describe the relationship between atoms, matter and elements, using the United States, which of the following statements is true?

Small Large Medium

- a. The whole US would be atoms, each state would be elements and the cities in that state would be matter. *-X*
- b. The whole US would be matter, each state would be elements and the cities in that state would be atoms. *E L M S*
- c. The whole US would be elements, each state would be atoms and the cities in that state would be matter.
- d. The whole US would be matter, each state would be atoms and the cities in that state would be elements.

Introduction To Matter Answer Key Person

Thomas Griffiths



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