

Lab 22A

Investigating Chemical Equilibrium

Purpose: To investigate the macroscopic properties of two chemical systems at equilibrium, observe shifts in equilibrium concentrations as a function of concentration and temperature and relate the the systems, which are exploring the structures through applying Le Chatelier's principle.

Proble:

1. Write a balanced equation for the reaction in part (a).
2. Write a balanced equation for the reaction in part (b).
3. Write a balanced equation for the reaction in part (c).
4. In part (a), what is the conjugate acid of the base? In part (b), what is the conjugate acid of the base? In part (c), what is the conjugate acid of the base? In part (d), what is the conjugate acid of the base? In part (e), what is the conjugate acid of the base? In part (f), what is the conjugate acid of the base? In part (g), what is the conjugate acid of the base? In part (h), what is the conjugate acid of the base? In part (i), what is the conjugate acid of the base? In part (j), what is the conjugate acid of the base? In part (k), what is the conjugate acid of the base? In part (l), what is the conjugate acid of the base? In part (m), what is the conjugate acid of the base? In part (n), what is the conjugate acid of the base? In part (o), what is the conjugate acid of the base? In part (p), what is the conjugate acid of the base? In part (q), what is the conjugate acid of the base? In part (r), what is the conjugate acid of the base? In part (s), what is the conjugate acid of the base? In part (t), what is the conjugate acid of the base? In part (u), what is the conjugate acid of the base? In part (v), what is the conjugate acid of the base? In part (w), what is the conjugate acid of the base? In part (x), what is the conjugate acid of the base? In part (y), what is the conjugate acid of the base? In part (z), what is the conjugate acid of the base?
5. Write the equation for the reaction between H^+ and OH^- ions.
6. Write the equation for the reaction between H^+ and OH^- ions.
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50. Write the equation for the reaction between H^+ and OH^- ions.

Investigating Chemical Equilibrium Lab 12a

LL Leslie



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