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Thomas Griffiths



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web solution a titrant volume 0 mL the solution ph is due to the acid ionization of HCl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the ph of the solution is then $\text{pH} = \log 0.100 = 1.000$ $\text{pH} = \log 0.100 = 1.000$ b titrant volume 12.50 mL

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4	6	4	6	5	9	5	9
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